

# Enhanced epitaxial growth of two-dimensional monolayer WS<sub>2</sub> film with large single domains

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## ABSTRACT

The emerging of graphene has stimulated the exploration of two-dimensional (2D) nanomaterials. As a kind of important semiconducting 2D materials, WS<sub>2</sub> shows great potentials in electronics, optoelectronics, and photonics, etc.; however, the synthesis of high quality, large area, uniform and epitaxial 2D WS<sub>2</sub> monolayer films is still a challenge. Herein, we report the epitaxial growth of WS<sub>2</sub> on sapphire with excellent electrical and optoelectronic properties via an enhanced chemical vapor deposition method. With the assistance of Na<sub>2</sub>WO<sub>4</sub> as well as the investigation of related growth mechanism, large single crystal triangular WS<sub>2</sub> monolayer flakes with well-defined orientations are achieved. The corresponding single crystal domain size is larger than 1 mm together with the centimeter-scale continuous film realized. Once fabricated into transistors, the monolayers exhibit excellent device properties, such as high mobility and large on/off current ratio. Also, these devices give respectable photoresponse properties when operated in the photoconductive mode. The peak responsivity can reach 4.6 A W<sup>-1</sup>. All these results demonstrate the high quality of as-synthesized WS<sub>2</sub> monolayers, indicating the bright future of this synthesis technology for 2D materials towards practical applications.

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## 1. Introduction

Due to the fascinating properties, two-dimensional (2D) nanomaterials with the atomic thickness have been widely demonstrated with excellent electrical and photoelectric characteristics for promising utilization in electronics, and optoelectronics, etc. [1–4]. Since there are a broad range of materials within the 2D family, including insulators, semiconductors, semimetals, and superconductors [5], different 2D monolayers can be heterogeneously integrated as both active and passive materials to build complete sets of devices, where their ultra-thin natures are advantageous for flexible and wearable applications [6]. Even compared to silicon, the “workhorse” of the electronic industry, 2D monolayers, such as MoS<sub>2</sub> and WS<sub>2</sub>, exhibit superior semiconducting properties [2,3], being capable to serve as ideal alternative active channel materials in the post-Si era.

For practical applications, the use of 2D materials inevitably requires reliable synthesis methods, especially to achieve the large-area, uniform, crystalline and epitaxial monolayer films; however, it is still a great challenge to realize these demanding synthesis techniques [7,8]. Among various synthesis schemes, chemical vapor deposition (CVD) is one of the most promising and cost-effective schemes to obtain large-scale and high-quality 2D materials [8,9]. In fact, during the past decade, there is a lot of pioneer works carried out towards the large-area and high-quality synthesis of 2D materials by different CVD methods. Particularly, Xu and his team realized the meter-sized single-crystalline graphene on industrial copper foils employing the CVD method [10]. Yu et al. reported the synthesis of large-area highly-oriented MoS<sub>2</sub> monolayer films by the CVD technique [11]. Later, the same group achieved the wafer-scale MoS<sub>2</sub> monolayers with large single crystal domains employing the modified CVD method [12]. In contrast to MoS<sub>2</sub>, WS<sub>2</sub> is another important semiconducting transition metal dichalcogenide 2D materials, but its large-scale epitaxial synthesis is still far lack behind although there are recent advances in various growth methods [13]. For example, Zhou and his group ex-

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explored the fast growth of WS<sub>2</sub> flakes (*i.e.* not necessarily monolayers) with the single crystal size reaching 500 μm using K<sub>2</sub>WS<sub>4</sub> as the growth precursor [14]. Gao et al. revealed that the very large single crystal size of WS<sub>2</sub> up to 1 mm could be potentially achieved by simply using Au foils as substrates [15]. Ji and his team found that the introduction of hydrogen could facilitate the epitaxial growth of monolayer WS<sub>2</sub> triangles on c-oriented sapphire substrates owing to the increased interaction between WS<sub>2</sub> and sapphire [16]. Nonetheless, the obtained single crystal size is relatively small with the lateral dimension of only around 3 μm, which significantly hinders its use for constructing large-scale devices. Epitaxial WS<sub>2</sub> monolayers with low in-plane rotational twist was lately achieved by Chubarov et al. using metalorganic CVD method with a multistep growth process [17]. Although unidirectional WS<sub>2</sub> monolayers have been achieved, the single domain WS<sub>2</sub> flakes are very small (~200 nm), resulting in large numbers of grain boundaries. Thus, epitaxial WS<sub>2</sub> monolayers with large single domains are highly preferred.

It has been reported that several inorganic salts, such as NaCl and KCl, can be exploited to reduce the melting point of metal oxides, facilitating the evaporation of metal source, and realizing the synthesis of various kinds of 2D materials [18]. In any case, it is difficult to have a rational control on the evaporation rate of metal oxides, which always leads to the formation of thick 2D materials. In this work, we demonstrate the successful synthesis of well-oriented large single domain size of WS<sub>2</sub> monolayer films by a modified CVD method. In the synthesis, mixed powders of WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O were employed as the W source. With the assistance of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O, the evaporation of W source can be well controlled, leading to the formation of large monolayer single crystal WS<sub>2</sub> triangular grains and subsequently uniform WS<sub>2</sub> monolayer films. Importantly, the as-synthesized single crystal triangular grains have well-aligned orientations to ensure the in-plane texture uniformity. When fabricated into simple back-gated devices, the obtained WS<sub>2</sub> monolayers exhibit decent electrical and photodetection performance, illustrating their promising applications in next-generation electronics and optoelectronics.

## 2. Materials and methods

### 2.1. Materials synthesis

For the synthesis of WS<sub>2</sub> monolayer films, a single heating zone furnace was utilized. The details of the setup can be found in Supporting Information Figure S1. To be specific, a quartz tube with diameter of 1 inch was used as the reaction chamber for the material growth. Mixed powders (800 mg) of WO<sub>3</sub> (99.8%, Xiya Reagent) and Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O (99.5%, Macklin) with different mass ratios were exploited as the W source, where they were placed in a quartz boat. The quartz boat was then positioned in the center of the heating zone. At the same time, 300 mg of S powder (99.95%, Aladdin) was employed as the S precursor, being placed in a corundum boat. The corundum boat was housed in the upstream side of the quartz tube, sitting outside of the heating zone. In this case, a heating belt was used to heat the S powder during the reaction while the temperature of the heating belt was set to be 150 °C. Several pieces of single side polished c-oriented sapphire were applied as the substrates, which were placed in the downstream side of heating zone of the furnace. Before heating, the pressure inside the quartz tube was first pumped to 1 Pa by a mechanical pump. Next, high purity Ar gas (99.999%) was introduced into the tube with a flow rate of 200 sccm. During the reaction, the pressure inside the quartz tube was maintained at about 140 Pa monitored by a pressure gauge (901p-11040, mks). The furnace was heated to 930 °C in 30 min and kept at that temperature for 60 min. After

the reaction, the furnace was cooled to room temperature naturally under Ar flow.

### 2.2. Characterization

The thermogravimetry and differential scanning calorimetry analysis of the precursors are performed with a thermal analyzer (STA 449 F3 Jupiter, NETZSCH). After the synthesis, the obtained WS<sub>2</sub> films were inspected by an optical microscope (BX53, Olympus) and a scanning electron microscope (SEM, VEGA3 SBH, TESCAN). An atomic force microscope (AFM, FM-Nanoview 1000, FSM-Precision) was used to examine the thickness of the WS<sub>2</sub> film. The Raman and PL spectra of the WS<sub>2</sub> films were collected by a spectrometer (SR-500i-A-R, Andor) with a 532 nm excitation laser. The excitation power was fixed at 2 mW for the PL and Raman measurements. A fiber-spectrometer (BLK-C-SR, Stellar Net Inc) with deuterium halogen light source (SL5-CUV, Stellar Net Inc) was utilized to measure the absorbance of the WS<sub>2</sub> films. For the absorbance measurements, the WS<sub>2</sub> films were transferred onto a glass substrate using polystyrene as the protecting layer [19,20]. A transmission electron microscope (TEM, 200 kV, Tecnai G2 F20, FEI) was used to assess the crystal structure and crystallinity of the WS<sub>2</sub> films.

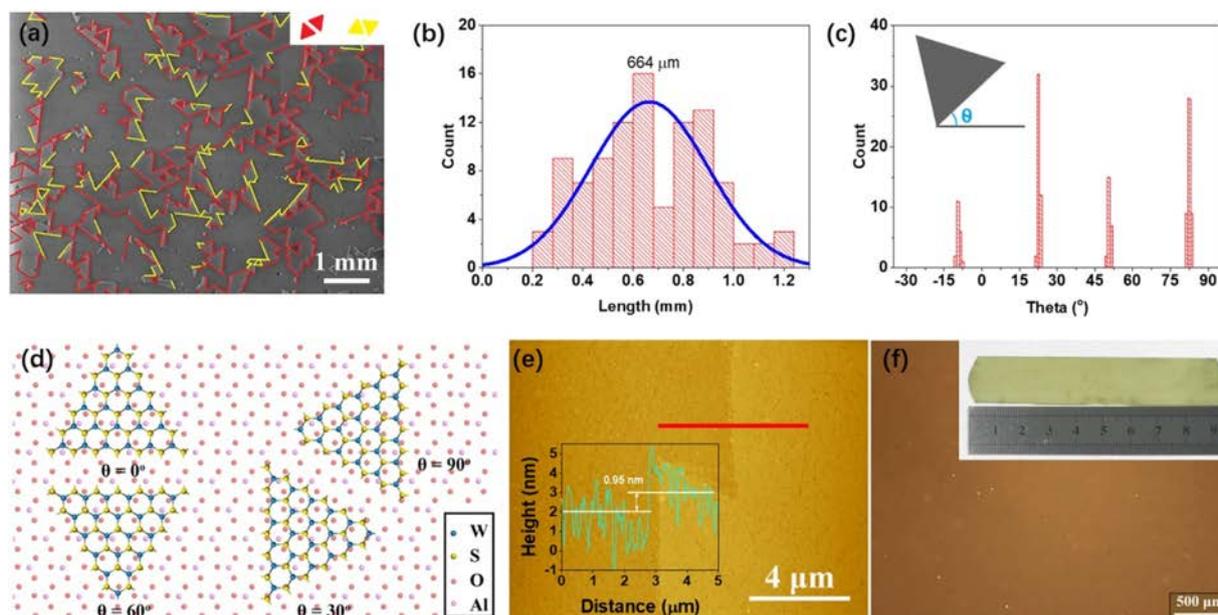
### 2.3. Devices fabrication and measurements

For the device fabrication, the as-synthesized WS<sub>2</sub> films were transferred from the sapphire substrate to the SiO<sub>2</sub>(285 nm)/Si substrate. After that, a standard lithography process was exploited to define the channel areas, which were protected by photoresist during the removal of other WS<sub>2</sub> regions by oxygen plasma. Another lithography process was then used to define the source and drain areas, followed by the deposition of Bi/Au stacks (5 nm/80 nm) using thermal evaporation and the lift-off process. The performance of fabricated back-gated field-effect transistors (FETs) was evaluated in a vacuum probe station with a pressure smaller than 1 × 10<sup>-3</sup> Pa. A semiconductor analyzer (B1500A, Keysight) was used to record the electrical data. The photodetection performance was assessed under ambient conditions. For the photodetection performance measurement, a 532 nm laser was employed, where its power was measured by an optical power meter (PM320E, Thorlabs) equipped with a Si diode photodetector (S130VC, Thorlabs).

## 3. Results and discussion

### 3.1. Morphology characterization

Here, (0001) oriented single side polished sapphire substrates were utilized for the CVD synthesis of WS<sub>2</sub>. Due to the similar lattice symmetry between WS<sub>2</sub> and sapphire, it is feasible to realize epitaxial growth although they interact with each other by the weak van der Waals interaction. Under the optimal synthesis condition (*i.e.* with weight ratio of WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O as 1:2, and reaction time of 40 min), large WS<sub>2</sub> triangular flakes with well-aligned orientations can be achieved as shown in Fig. 1a. Their size distribution is then extracted from Fig. 1a and depicted in Fig. 1b. Based on the normal distribution fitting, a mean value of 664 μm is obtained, suggesting the relatively large size of these triangular flakes. In fact, the largest WS<sub>2</sub> flakes can exceed 1 mm (Figure S2, Supporting Information). The well-aligned triangular flakes can be clearly seen from Fig. 1a. In order to illustrate the aligned orientation more accurately, the orientation distribution is extracted from Fig. 1a and presented in Fig. 1c. It is obvious that the flakes show four dominant orientations, indicating the epitaxial growth of WS<sub>2</sub> on sapphire substrate. Such large single crys-



**Fig. 1.** Morphology, orientation and size distribution of the monolayer WS<sub>2</sub> flakes. (a) SEM image of the WS<sub>2</sub> flakes. The red lines and yellow lines indicate the WS<sub>2</sub> triangular grains with different orientations. Inset shows the schematic of the triangular grains with different orientations. (b) Size distribution of the triangular grains. (c) Orientation distribution of the WS<sub>2</sub> triangular grains. Inset illustrates the schematic of the determination of the theta value. (d) Schematic of the orientation relationship between WS<sub>2</sub> and sapphire substrate. (e) AFM image of a typical monolayer WS<sub>2</sub> flake. Inset presents the height profile along the edge indicated by the red line. (f) Optical microscope image of a continuous WS<sub>2</sub> film. Inset gives the digital photograph of a continuous WS<sub>2</sub> film.

tals with well-aligned orientations are reported for the first time to the best of our knowledge. The most probable relative orientation for the monolayer WS<sub>2</sub> grown on sapphire surface is schematically displayed in Fig. 1d. The 0° and 60° orientations correspond to the 3-on-2 superstructure of WS<sub>2</sub> and sapphire because the distance between adjacent S atoms of WS<sub>2</sub> is 3.18 Å and the distance between adjacent O atoms of Al<sub>2</sub>O<sub>3</sub> is 4.81 Å. The similar phenomenon was also observed in the epitaxial growth of MoS<sub>2</sub> on sapphire [21,22]. On the other hand, there is not any strict lattice match for the other two orientations (*i.e.* 30° and 90°), which have similar atom alignments with the sapphire substrate. For determining the thickness of triangles flakes, AFM measurement was carried out. A typical AFM image of the WS<sub>2</sub> flake is shown in Fig. 1e. The WS<sub>2</sub> flake shows a thickness of 0.95 nm, corresponding to the monolayer of WS<sub>2</sub>. For obtaining a continuous film, the longer growth time is implemented. In our experiments, a growth duration of 60 min is optimal for the growth of a continuous film (Fig. 1f). Remarkably, the centimeter scale continuous monolayer WS<sub>2</sub> film can be achieved, where the size of 1.7 × 9.5 cm<sup>2</sup> is displayed in the inset of Fig. 1f. It is anticipated that the wafer scale monolayer WS<sub>2</sub> film can be realized when a larger tube furnace is used.

### 3.2. Microstructure and crystallinity characterization

In order to evaluate the crystallinity of obtained WS<sub>2</sub> monolayer films, thorough TEM measurements were performed. A representative TEM image is shown in Fig. 2a. The observed cracks and triangular holes may be generated during the sample preparation. Notably, the uniform contrast of the film suggests the good uniformity of the grown film. High-resolution TEM images were also recorded to examine the crystal structure and crystallinity of the WS<sub>2</sub> film with the typical image given in Fig. 2b. It is evident that clear lattice fringes can be seen, designating the good crystallinity of the WS<sub>2</sub>. Lattice spacings of 0.27 nm is observed (Fig. 2b), which corresponds to the spacing of the {100} planes of WS<sub>2</sub>. It should also be noticed that during the HRTEM measurements, the WS<sub>2</sub> crystal

could be destroyed due to the strong electron-beam irradiation. In this way, the cracks shown in the image are most probably caused by the electron-beam damage. The high crystallinity of WS<sub>2</sub> can be confirmed by the selected area electron diffraction (SAED) measurements. As depicted in the typical SAED pattern in Fig. 2c, there is only one set of diffraction spots observed, indicating the single crystallinity of the detected area with the diffraction spots assigned to the {100} planes of WS<sub>2</sub>. All these results can clearly reveal the excellent crystallinity of our WS<sub>2</sub> monolayer films.

### 3.3. Optical spectra characterization

After that, different optical spectra were collected to further characterize the as-synthesized WS<sub>2</sub> monolayer films. A typical Raman spectrum with characteristic peaks labelled corresponding to WS<sub>2</sub> is displayed in Fig. 3a. These sharp Raman peaks indicate the good crystallinity of the film, being consistent with the TEM results discussed above. Because of the close positions of the 2LA(M) – E<sub>2g</sub><sup>2</sup>(Γ), E<sub>2g</sub><sup>1</sup>(M), 2LA(M), and E<sub>2g</sub><sup>1</sup>(Γ) Raman modes, Lorentzian fitting is used to determine to exact peak positions as compiled and shown in Fig. 3a. In this way, the energy difference between E<sub>2g</sub><sup>1</sup>(Γ) and A<sub>1g</sub>(Γ) peaks is determined to be 59.3 cm<sup>-1</sup>, signifying that the WS<sub>2</sub> film is indeed a monolayer film [23]. The strong 2LA(M) peak caused by the double resonance scattering is much stronger than that of the A<sub>1g</sub>(Γ) peak in monolayer WS<sub>2</sub>, further confirming the single layer of the WS<sub>2</sub> film [24]. Since monolayer WS<sub>2</sub> is a direct bandgap semiconductor, it should have strong photoluminescence (PL) peaks. Based on the collected PL signals as given in Fig. 3b, a strong PL peak is observed, again revealing that the obtained WS<sub>2</sub> is a single layer film. It is worth mentioning that the PL spectrum can be deconvoluted into four sub-peaks according to Lorentzian fitting, which are located at 1.884, 1.933, 1.977, and 2.006 eV, respectively. The peak located at 1.884 eV is attributed to the defects while the peak of 1.933 eV is ascribed to the bi-exciton/defect [25]. The peaks located at 1.977 and 2.006 eV are belonged to the signals of trion and exciton of monolayer WS<sub>2</sub>, respectively [26,27]. After transferring the WS<sub>2</sub> film onto the SiO<sub>2</sub>/Si

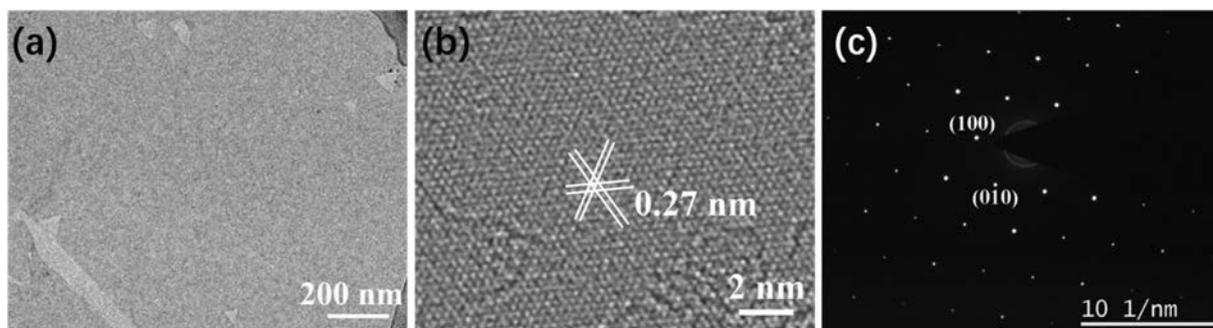


Fig. 2. TEM characterization of the obtained WS<sub>2</sub>. (a) Low-magnified TEM image. (b) High-resolution TEM image. (c) SAED pattern.

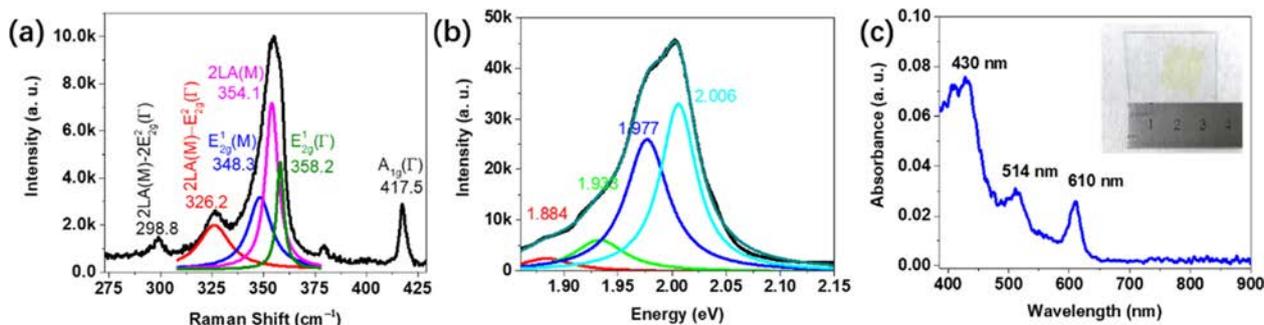


Fig. 3. Optical spectra of the monolayer WS<sub>2</sub>. (a) Raman spectrum. The peaks are fitted by Lorentz function. (b) PL spectrum. The peaks are fitted by Lorentz function. (c) Absorbance spectrum. Inset shows the monolayer WS<sub>2</sub> film transferred onto a glass substrate. For Raman and PL measurements, a 532 nm laser was used as the excitation source.

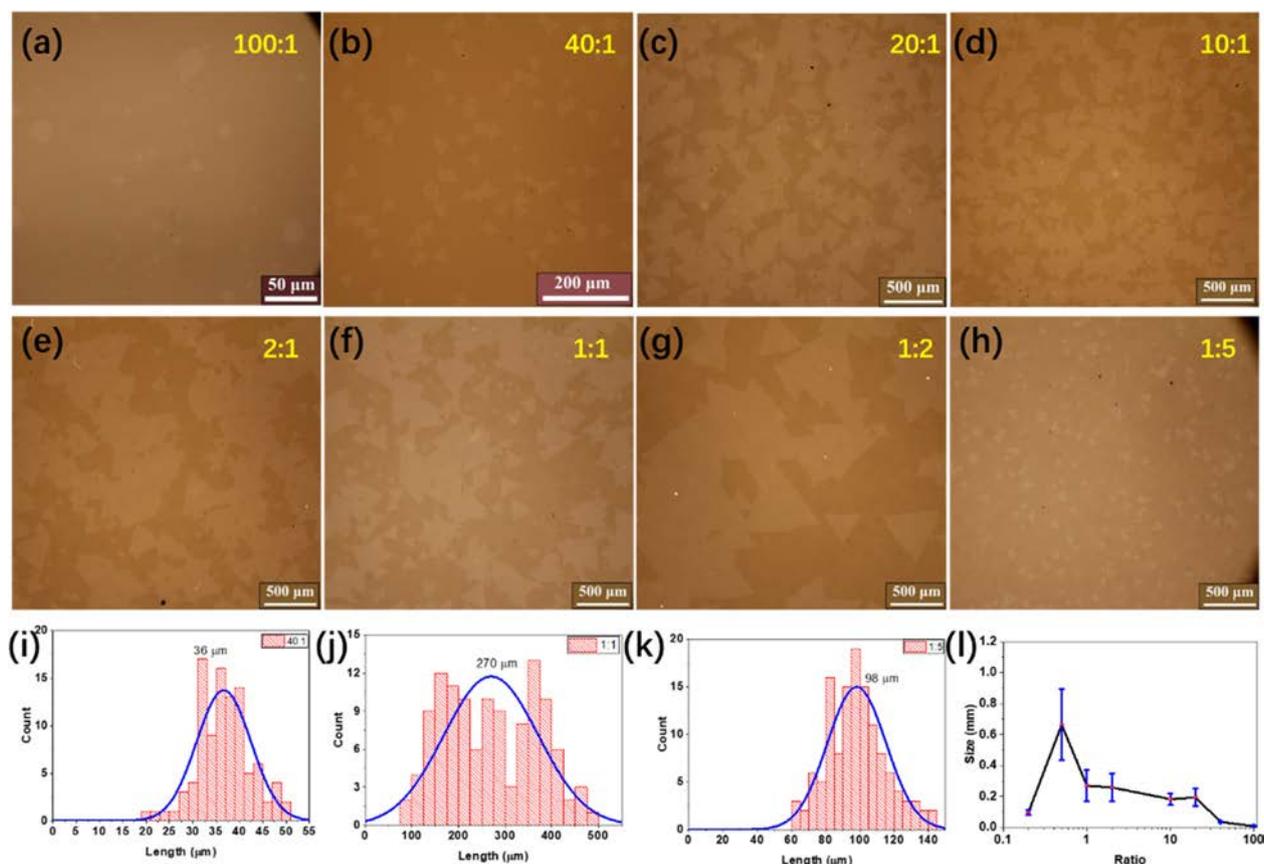
substrate, the strain in WS<sub>2</sub> can be released, resulting in a blue shift of the PL peak (Figure S3, Supporting Information) [28]. As large-area WS<sub>2</sub> films were obtained here, it is convenient to measure the absorbance spectrum after the transfer of WS<sub>2</sub> onto the glass substrate as shown in the inset of Fig. 3c. The corresponding absorbance spectrum is presented in Fig. 3c. It is clear that there are three excitonic peaks observed that are positioned at 610, 514, and 430 nm. These absorption peaks are attributed to A, B, and C excitons in the monolayer WS<sub>2</sub>, respectively [29]. Peak A is linked to the excitonic absorption in direct gap of WS<sub>2</sub> films, where it is located at the K valley of the Brillouin zone [30]. Therefore, the excitonic A peak should be correlated with the PL peak due to the same origin. The peak located at 610 nm is corresponded to the photon energy of 2.033 eV, which is perfectly consistent with the PL peak of monolayer WS<sub>2</sub> transferred onto the SiO<sub>2</sub>/Si substrate, indicating the same origin of the peak. In addition, the energy difference between the excitonic peaks A and C is highly sensitive to the thickness of WS<sub>2</sub>, in which it is usually used to determine the thickness of WS<sub>2</sub> films [29,30]. Based on the energy difference of 0.851 eV between these two excitonic peaks, it is again confirmed that the obtained WS<sub>2</sub> is a monolayer film [29].

### 3.4. Growth mechanism

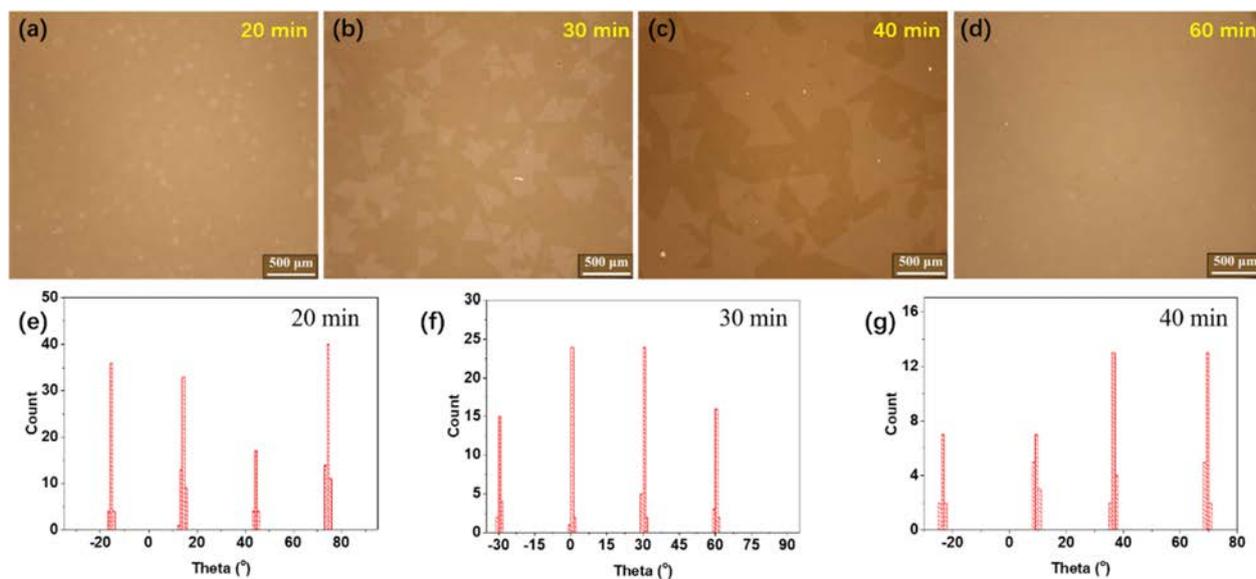
To shed light on the growth mechanism of these large single crystal monolayer WS<sub>2</sub> epitaxial films, the growth behavior of WS<sub>2</sub> with varied growth parameters were carefully investigated. Because adding Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O into the precursor is a key factor for the growth, the morphology evolution as a function of the weight ratio between WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O was examined with the results compiled in Fig. 4. When a small amount of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O was mixed with WO<sub>3</sub>, the growth became very slow with only a few dispersed small WS<sub>2</sub> flakes found on the substrate (Fig. 4a and 4b). In addition, the WS<sub>2</sub> flakes show the round-like shape when the weight ratio between WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub> is 100:1, which

may be due to the ultra-slow growth rate. With the increasing amount of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O, the morphology of WS<sub>2</sub> changed into triangles as displayed in Fig. 4b. However, these triangular flakes have random orientations as confirmed by the corresponding orientation distribution (Figure S4a, Supporting Information). Once the amount of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O continued to increase, large triangular flakes with well-aligned orientations are witnessed (Fig. 4c-4g) with the respective orientation distributions compiled in Supporting Information Figure S3. These results designate that the addition of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O is beneficial for facilitating the epitaxial growth of WS<sub>2</sub> films. The average size of triangular flakes is found to gradually increase with the decreasing weight ratio between WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub> (Fig. 4i-4l). However, the size of triangular flakes would start to decrease by further increasing the amount of Na<sub>2</sub>WO<sub>4</sub> (Fig. 4h, 4k and 4l) even though the triangular flakes still exhibit well-defined orientations (see the statistical orientations shown in Figure S4, Supporting Information). Furthermore, there would not be any WS<sub>2</sub> flakes produced without the addition of WO<sub>3</sub>. All these findings indicate that the amount of Na<sub>2</sub>WO<sub>4</sub>•2H<sub>2</sub>O is important for controlling the morphology evolution of WS<sub>2</sub> monolayer films.

Simultaneously, the time evolution of the morphology of WS<sub>2</sub> films were also cautiously studied as illustrated in Fig. 5. In the early state of the growth, only small triangular WS<sub>2</sub> flakes are observed on the substrate (Fig. 5a). These small-size flakes show the epitaxial relation with the underlying substrate as witnessed from the orientation distribution of these triangular flakes (Fig. 5e). With the increasing growth time, these small triangular flakes gradually grow into the larger flakes (Fig. 5b and 5c, Figure S5a-S5c, Supporting Information) and finally merge into a continuous film (Fig. 5d). It is impressive that these triangular flakes maintain their epitaxial relationship with the substrate during the growth (Fig. 5e-5g). In fact, the nucleation density is found to decrease with the increasing growth time (Figure S5d, Supporting Information). The reduced density of nuclei and increased size of triangular flakes designate that the growth of the monolayer WS<sub>2</sub> trian-



**Fig. 4.** Morphology and size evolution as a function of the weight ratio between  $\text{WO}_3$  and  $\text{Na}_2\text{WO}_4$ . Optical microscope image of the samples. (a) 100:1, (b) 40:1, (c) 20:1, (d) 8:1, (e) 2:1, (f) 1:1, (g) 1:2, (h) 1:5. Size distribution of the  $\text{WS}_2$  triangular flakes. (i) 40:1, (j) 1:1, (k) 1:5. (l) Size distribution of the flakes as a function of the weight ratio between  $\text{WO}_3$  and  $\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$ . The growth time is 40 min.



**Fig. 5.** Time evolution of the morphology of  $\text{WS}_2$  monolayer films with the weight ratio of  $\text{WO}_3$  and  $\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$  kept at 1:2. Optical microscope images of the samples. (a) 20 min, (b) 30 min, (c) 40 min, (d) 60 min. Orientation distributions of the samples. (e) 20 min, (f) 30 min, (g) 40 min.

gular flakes follows the mechanism of Ostwald ripening or Smoluchowski ripening [31].

Based on the above discussion, a growth mechanism can be proposed for the  $\text{Na}_2\text{WO}_4$ -assisted growth of  $\text{WS}_2$  monolayer films (Fig. 6). During the growth, it is anticipated that the mixture of  $\text{WO}_3$  and  $\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$  would first change into liquid at high

temperatures since  $\text{Na}_2\text{WO}_4$  has a relatively low melting point (698 °C). Then, there is a reaction occurred between  $\text{Na}_2\text{WO}_4$  and  $\text{WO}_3$ , leading to the formation of liquid  $\text{Na}_2\text{W}_x\text{O}_y$ , where  $x$  and  $y$  indicate the atomic content relative to two Na atoms. This way,  $\text{Na}_2\text{W}_x\text{O}_y$  would evaporate at high temperatures, delivering  $\text{Na}_2\text{W}_x\text{O}_y$  vapor. As the  $\text{Na}_2\text{W}_x\text{O}_y$  vapor is not stable, it will de-

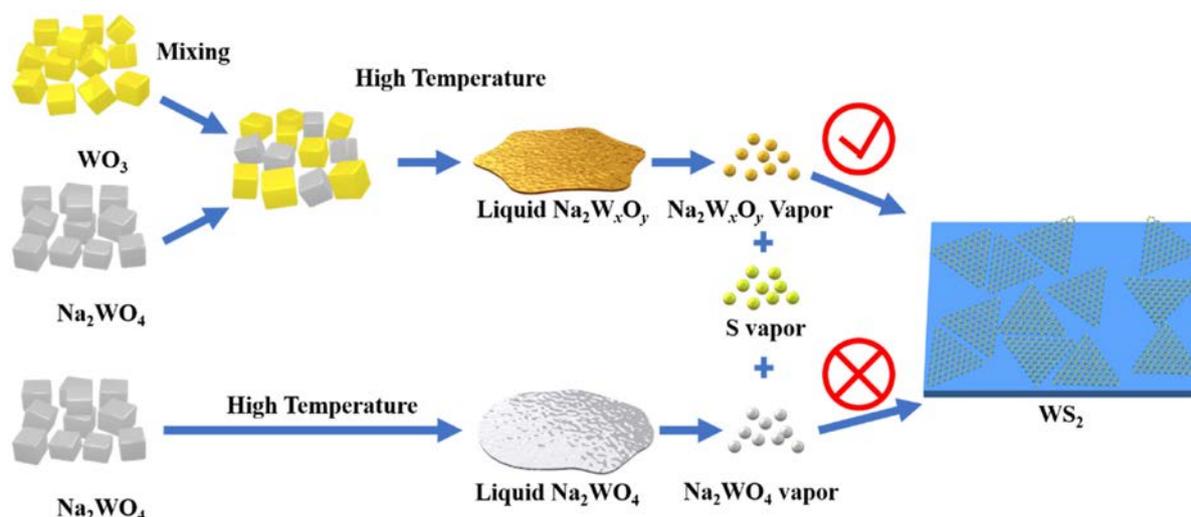


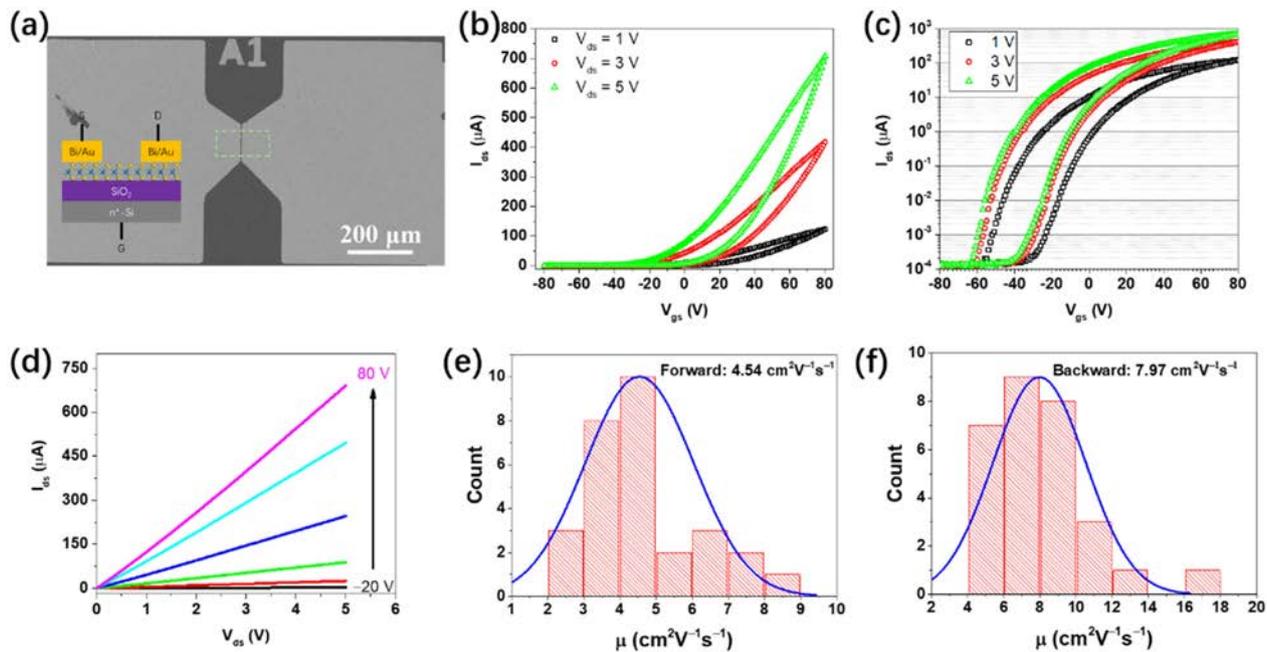
Fig. 6. Proposed growth mechanism of the monolayer WS<sub>2</sub> triangular flakes.

compose into Na<sub>2</sub>WO<sub>4</sub> and WO<sub>2</sub>. This hypothesis is supported by the thermogravimetry and differential scanning calorimetry analysis (Figure S6, Supporting Information). These newly formed WO<sub>2</sub> species have the high chemical activity, which can react with S easily, forming WS<sub>2</sub> species and transporting to the substrate by the carrier gas. Once they hit the substrate, WS<sub>2</sub> seeds can be nucleated to form the triangular flakes. Due to the similar lattice symmetry between WS<sub>2</sub> and c-oriented sapphire, there is a possibility for the formation of van der Waals epitaxy, yielding well-aligned triangular WS<sub>2</sub> flakes on the substrate. As a result, the Na<sub>2</sub>WO<sub>4</sub> species may have four functions here. Firstly, they can facilitate the continuous and stable feeding of WO<sub>2</sub> species during the reaction. This is because the newly formed Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> liquid does not react with S directly. In fact, after the reaction, there is not any WS<sub>2</sub> but Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> observed in the precursor boat (Figure S7, Supporting Information). Furthermore, the Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> species can still be used for the synthesis of WS<sub>2</sub>, which has been confirmed in our experiments, suggesting the validity of the hypothesis. Also, the Na<sub>2</sub>WO<sub>4</sub> species cannot release WO<sub>2</sub> because no WS<sub>2</sub> can be found on the substrate when using only Na<sub>2</sub>WO<sub>4</sub> as the W source. The Na<sub>2</sub>WO<sub>4</sub> precursor still remains as pure Na<sub>2</sub>WO<sub>4</sub> after the reaction (Figure S8, Supporting Information). Secondly, the Na<sub>2</sub>WO<sub>4</sub> species can provide homogeneous WO<sub>2</sub> during the growth due to the reaction between Na<sub>2</sub>WO<sub>4</sub> and WO<sub>3</sub>. The evaporation of liquid Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> is a more stable process as compared to that of its solid powder. Thus, homogeneous Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> can be formed at high temperatures, which leads to the formation of homogeneous WO<sub>2</sub> species. Thirdly, the existence of Na species may reduce the reaction barrier between WO<sub>2</sub> and S, yielding the fast formation of WS<sub>2</sub> species, which has been reported for the growth of MoS<sub>2</sub> [32]. Fourthly, the Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> species may increase the interaction between WS<sub>2</sub> and sapphire, leading to the epitaxial growth of WS<sub>2</sub> on sapphire. Actually, the quasi-epitaxy between triangular WS<sub>2</sub> flakes and sapphire has been witnessed in our previous work using NaOH as a promoter [26], but the exact mechanism behind the phenomenon still needs further studies. In this case, the morphology evolution of WS<sub>2</sub> on the weight ratio between WO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub>·2H<sub>2</sub>O can be explained as the following. When the amount of Na<sub>2</sub>WO<sub>4</sub> is low, the amount of WO<sub>2</sub> species generated is low, inducing the slow growth speed and low growth yield of WS<sub>2</sub>. In addition, the amount of Na species is relatively low in the region near the substrate, which is not sufficient for guiding the epitaxial growth of WS<sub>2</sub> on sapphire. As a result, randomly distributed triangular WS<sub>2</sub> flakes are formed. With the gradually increasing amount

of Na<sub>2</sub>WO<sub>4</sub>, the generated WO<sub>2</sub> species would increase drastically, giving the dense nucleation and fast growth of the WS<sub>2</sub> flakes. Therefore, the dense triangular flakes with well-aligned orientations are obtained. The release of WO<sub>2</sub> species should be similar for a wide range of weight ratio (e.g. 20 to 1) because the triangular flakes have the similar size as indicated in Fig. 4. Further increasing the amount of Na<sub>2</sub>WO<sub>4</sub> can reduce the amount of WO<sub>2</sub> species as the index of y in Na<sub>2</sub>W<sub>x</sub>O<sub>y</sub> is close to 1. This way, once the nucleation density is reduced, the larger triangular WS<sub>2</sub> flakes are formed. When the amount of Na<sub>2</sub>WO<sub>4</sub> increases too much, the release amount of WO<sub>2</sub> species is drastically reduced. As a result, the growth rate of WS<sub>2</sub> is reduced with the formation of small WS<sub>2</sub> flakes. Recently, it was reported that graphene nanoribbons can be grown with two different edges by tuning the orientations of the nano-trenches on h-BN substrate [33]. However, due to the large lattice mismatch between WS<sub>2</sub> (3.18 Å) and h-BN (2.51 Å), it may be impossible to have a similar template growth behavior for WS<sub>2</sub>. While due to the similar lattice symmetry, it is possible to realize van der Waal epitaxial growth of WS<sub>2</sub> monolayers on h-BN [34]. Further work is needed towards the growth of unidirectional large single domain WS<sub>2</sub> monolayers.

### 3.5. Electrical properties

Since WS<sub>2</sub> is considered as a potential candidate active material for next generation electronics, it is necessary to explore the electrical properties of as-synthesized WS<sub>2</sub> films. Here, global back-gated FETs are constructed using the monolayer WS<sub>2</sub> film as device channel as schematically presented in the inset of Fig. 7a. The SEM image of a representative FET is shown in Fig. 7a, which has a channel length (*L*) of 3.7 μm and width (*W*) of 79 μm. Because it has been reported that Bi contact is beneficial for the formation of ohmic contact between WS<sub>2</sub> and Bi [35], Bi contact is employed for our devices. The FET shows the typical n-type semiconducting properties as it can be seen from the transfer curves as shown in Fig. 7b and 7c, where the current increases with the increasing gate voltage. The on/off current ratio can then be determined from the logarithmic current plot (Fig. 7c), in which it is larger than 10<sup>6</sup>, being comparable to the values reported in literatures (Table 1). The large hysteresis in the transfer curves may be caused by charge traps arising from various origins, such as the interface states existed between WS<sub>2</sub> channel and SiO<sub>2</sub> dielectric, defects in WS<sub>2</sub>, and charge traps in the SiO<sub>2</sub> dielectric, etc. [36–39]. The corresponding output curves is given in Fig. 7d. The current in-



**Fig. 7.** Electrical properties of the obtained WS<sub>2</sub> films. (a) SEM image of a typical FET. The green dashed square indicates the WS<sub>2</sub> channel region. Inset shows the device schematic of the FET. (b) Transfer curves. (c) Transfer curves with the logarithmic current axis. (d) Output curves. (e) Statistical of mobility calculated from the transfer curve in the forward sweep. (f) Statistical of mobility calculated from the transfer curve in the backward sweep.

**Table 1**  
Electrical properties of monolayer WS<sub>2</sub> films synthesized by different CVD methods.

Single domain size( $\mu\text{m}$ )	Mobility( $\text{cm}^2 \bullet \text{V}^{-1} \bullet \text{s}^{-1}$ )	On/off current ratio	Maximum current density( $\mu\text{A}/\mu\text{m}$ )	Ref.
~40	6	$10^6$	35	[40]
~500	4 – 14	$10^7 - 10^8$	~0.15	[14]
~233	32.3	$10^6$	7.3	[41]
Mean: 576 Maximum: ~1000	1–2	~ $10^7$	~2	[15]
~420	20	~ $10^8$	~1	[42]
300 – 400	~6	~ $10^5$	~0.24	[43]
Maximum:	Forward maximum	$>10^6$	8.7	Our
1200	Forward mean			work
Mean:	Backward maximum			
664	Backward mean			

creases with the increasing gate voltage, being consistent with the transfer curves. The linear output curves ( $I_{ds}$ - $V_{ds}$  curves) suggest the ohmic-like electrical contact formed between metal electrodes and device channels for the efficient carrier transport and extraction. Thus, an on current density of  $8.7 \mu\text{A}/\mu\text{m}$  is obtained, which is considered to be relatively high among different state-of-the-art WS<sub>2</sub> devices (Table 1). Based on the output curves, it can be inferred that the device operates in the linear regime. In this regard, the mobility of the FETs can be estimated using the following analytical equation:

$$\mu = \frac{\partial I_{ds}}{\partial V_{gs}} \cdot \frac{L}{W} \cdot \frac{1}{C_{ox} V_{ds}} \quad (1)$$

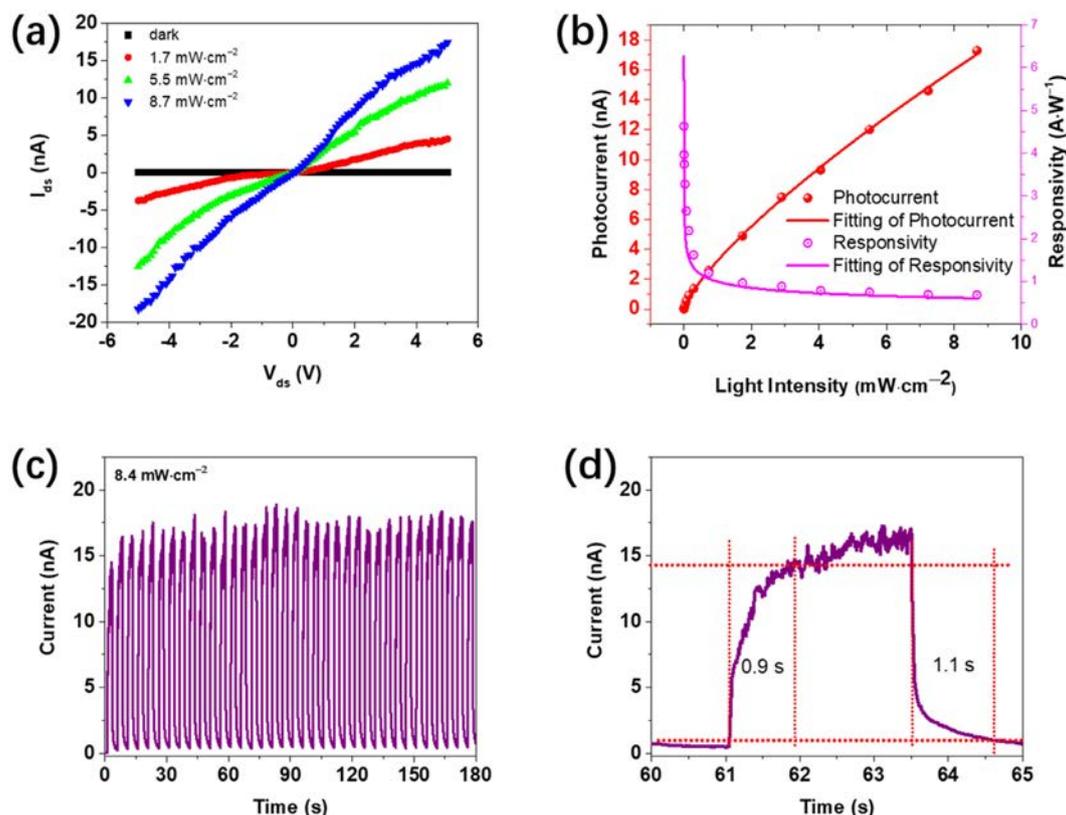
where  $C_{ox}$  is the gate capacitance per unit area (i.e.  $1.21 \times 10^{-4} \text{ F m}^{-2}$  for 285 nm thick SiO<sub>2</sub>). In this way, the mobility of the device demonstrated in Fig. 7b is calculated to be 7.0 and  $12.4 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$  for the forward and backward sweepings, respectively. To assess the uniformity of the electrical performance of the WS<sub>2</sub> film, tens of FETs were configured and the statistics of the extracted mobility of these FETs are compiled in Fig. 7e and 7f. The mean mobility is  $4.54 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$  for the forward sweeping, while it increases to  $7.97 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$  for the backward sweeping. These values are already comparable with the ones of CVD-synthesized monolayer WS<sub>2</sub> films reported in other recent works (Table 1), suggesting the good crystal quality of our WS<sub>2</sub> films.

### 3.6. Optoelectronic properties

Apart from FETs, WS<sub>2</sub> is also a good optoelectronic material with promising applications for optoelectronic devices. In this work, the photodetection performance of the obtained WS<sub>2</sub> films is explored. The  $I_{ds}$ - $V_{ds}$  curves with and without light irradiation of the device is shown in Fig. 8a. It is obvious that the device current increases with the increasing light intensity, indicating the good photosensitivity of WS<sub>2</sub>. In order to have a deeper understanding of the device characteristics, photocurrent ( $I_p$ ) vs. light intensity ( $\Phi$ ) was measured and is depicted in Fig. 8b. It is seen that the photocurrent increases with the increasing light intensity, which is perfectly consistent with the  $I_{ds}$ - $V_{ds}$  curves as shown in Fig. 8a. The relationship between photocurrent and light intensity can be well fitted by a sublinear equation of

$$I_p = A\Phi^\alpha \quad (2)$$

where  $A$  and  $\alpha$  are the fitting parameters. Based on the fitting from the measured data,  $\alpha$  is found to be 0.77. This sublinear relationship of the photocurrent with respect to light intensity is caused by the complex generation, trapping, and recombination processes of the photogenerated carriers, which are always observed in semiconductors based photoconductive detectors [44–46]. Moreover, responsivity is an important figure of merit for evaluating the perfor-



**Fig. 8.** Optoelectronic properties of the obtained WS<sub>2</sub> films. (a) Current-voltage curves without and with the light illumination. (b) Photocurrent and responsivity as a function of light intensity. (c) Time-dependent current under the chopped light illumination. (d) Time-dependent current showing the determination of rise and decay time constants. The wavelength of the light illumination is 532 nm.

mance of a photodetector, which is defined as

$$R = \frac{I_p}{\Phi S} \quad (3)$$

where  $S$  is the sensitive area of the photodetector (*i.e.* 292.3  $\mu\text{m}^2$  for the studied device as shown in Fig. 8). The calculated  $R$  as a function of light intensity is displayed in Fig. 8b. According to Eq. (2) and (3), the responsivity  $R$  should be proportional to  $\Phi^{\alpha-1}$  that is proportional to  $\Phi^{-0.23}$ . Thus, the responsivity decreases with the increasing light intensity. A maximum responsivity of 4.6  $\text{A W}^{-1}$  is obtained when the light intensity is  $3.3 \times 10^{-4} \text{ mW cm}^{-2}$ . With the increasing light intensity, the responsivity would reach a saturation value of 0.68  $\text{A W}^{-1}$ . This high responsivity at the low light intensity is caused by charge traps, which are frequently observed in low-dimensional materials based photoconductive detectors [47–49]. At this low light intensity, photogenerated holes are trapped by defects, the recombination rate of photogenerated electrons is reduced, leading to the increase of photogenerated electron lifetime ( $\tau$ ). Also, the gain ( $G$ ) of a photoconductive detector is known as  $G = \tau/t_{tr}$ , where  $t_{tr}$  is the transit time of electrons between electrodes. Therefore, the gain would increase at the low light intensity, leading to the increase of responsivity. At the high light intensity, most of the traps are filled by photogenerated holes, resulting in the increased recombination rate of photogenerated electrons. In this way, the responsivity reduces accordingly. Another important parameter is specific detectivity ( $D^*$ ), which can be written as  $D^* = R \cdot (S/2qI_{\text{dark}})$  if we consider the main noise is from shot noise here,  $q$  is the charge of an electron while  $I_{\text{dark}}$  is the dark current of the photodetector. According to the formula,  $D^*$  is proportional to  $R$ , where the highest  $D^*$  is determined to be  $2.2 \times 10^{12}$  Jones. The  $D^*$  will reach a saturation value of  $3.3 \times 10^{12}$  Jones under the high light intensity (Fig-

ure S9, Supporting Information). More importantly, the operation stability of a photodetector under modulated light irradiation is essential for practical applications. As shown in Fig. 8c, the stable on- and off-state current under modulated light illumination are witnessed, confirming the excellent stability of the device. The high-resolution current-time curve is as well shown in Fig. 8d, which is used to determine the response speed of the photodetector. The rise and decay time constants are defined as the time interval for photocurrent varied from 10 to 90% and vice versa, respectively. Accordingly, the rise and decay time constants are found to be 0.9 and 1.1 s, respectively. These photodetection parameters are already comparable or even better than the values reported in literatures (Table 2). This respectable photodetection performance clearly suggests the potential of our monolayer WS<sub>2</sub> films for next-generation optoelectronics.

#### 4. Conclusion

In summary, epitaxial monolayer WS<sub>2</sub> films with ultra-large single crystal domains are successfully synthesized by the modified CVD method. These high quality WS<sub>2</sub> monolayer films are characterized by optical microscopy, SEM, TEM, and optical spectroscopies in details. It is found that the introduction of Na<sub>2</sub>WO<sub>4</sub> in the precursor is extremely important to achieve large single domains of WS<sub>2</sub> with well-aligned epitaxial relationship with underlying sapphire substrates. Importantly, when configured into back-gated FETs, they demonstrate large electron mobilities with a mean mobility of 4.54  $\text{cm}^2\text{V}^{-1}\text{s}^{-1}$  for the forward sweeping and a mean mobility of 7.97  $\text{cm}^2\text{V}^{-1}\text{s}^{-1}$  for the backward sweeping, being comparable to other state-of-the-art CVD grown WS<sub>2</sub> devices. These devices also exhibit respectable photodetection performance with a maximum photoresponsivity of 4.7  $\text{A W}^{-1}$ . All these results sug-

**Table 2**  
Performance Parameters of photodetectors based on monolayer WS<sub>2</sub>.

Electrode	V <sub>ds</sub> (V)	V <sub>gs</sub> (V)	Channel length (μm)	Environment	Wavelength (nm)	R (A W <sup>-1</sup> )	D* (Jones)	Speed (s)	Ref.
Graphene	5	0	0.2	Ambient	532	2.5	9.9 × 10 <sup>10</sup>	–	[50]
Au	4	0	0.2	Ambient	532	20	–	–	[51]
Graphene	5	30	1	Ambient	532	121	–	–	[52]
Ti/Au	2	0	2	Vacuum	532	3.2 × 10 <sup>-3</sup>	–	0.17	[26]
Bi/Au	5	0	3.7	Ambient	532	4.6	2.2 × 10 <sup>12</sup>	1.1	This work

gest that the modified CVD method here does not only provide a facile platform for the synthesis of high quality, large single crystal, epitaxial monolayer WS<sub>2</sub> films, but also facilitates their excellent intrinsic materials properties for next-generation electronics and optoelectronics.

### Appendix. Supplementary data

Supplementary data associated with this article can be found, in the online version, at

### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

### CRediT authorship contribution statement

**Changyong Lan:** Conceptualization, Methodology, Investigation, Validation, Writing – original draft, Writing – review & editing, Supervision, Funding acquisition. **Rui Zhang:** Investigation, Visualization. **Haolun Wu:** Investigation, Visualization. **Shaofeng Wen:** Investigation, Visualization. **Ruisen Zou:** Investigation. **Xiaolin Kang:** Investigation. **Chun Li:** Conceptualization, Supervision, Writing – review & editing, Funding acquisition. **Johnny C. Ho:** Conceptualization, Supervision, Writing – review & editing, Funding acquisition. **Yi Yin:** Writing – review & editing. **Yong Liu:** Writing – review & editing.

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### Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.apmt.2021.101232.

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