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# Low-temperature growth of highly crystalline $\beta$ -Ga<sub>2</sub>O<sub>3</sub> nanowires by solid-source chemical vapor deposition

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# Abstract

Growing Ga<sub>2</sub>O<sub>3</sub> dielectric materials at a moderately low temperature is important for the further development of high-mobility III-V semiconductor-based nanoelectronics. Here,  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> nanowires are successfully synthesized at a relatively low temperature of 610°C by solid-source chemical vapor deposition employing GaAs powders as the source material, which is in a distinct contrast to the typical synthesis temperature of above 1,000°C as reported by other methods. In this work, the prepared  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> nanowires are mainly composed of Ga and O elements with an atomic ratio of approximately 2:3. Importantly, they are highly crystalline in the monoclinic structure with varied growth orientations in low-index planes. The bandgap of the  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> nanowires is determined to be 251 nm (approximately 4.94 eV), in good accordance with the literature. Also, electrical characterization reveals that the individual nanowire has a resistivity of up to  $8.5 \times 10^7 \Omega$  cm, when fabricated in the configuration of parallel arrays, further indicating the promise of growing these highly insulating Ga<sub>2</sub>O<sub>3</sub> materials in this III-V nanowire-compatible growth condition.

**Keywords:**  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> nanowires; Chemical vapor deposition; Solid-source; Highly crystalline; Large resistance; Dielectric

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# Background

In the past decade, gallium oxide (Ga<sub>2</sub>O<sub>3</sub>), as a largebandgap (approximately 4.9 eV) semiconductor, has attracted extensive attention in the area of insulating oxides for the metal-oxide-semiconductor (MOS) technology as well as the active materials for the solar-blind deep ultraviolet detectors [1-6]. In particular, when high-mobility III-V compound semiconductor nanomaterials, such as GaAs, InAs, GaSb, and InSb nanowires (NWs), have been successfully illustrated with their great technological potentials in next-generation electronics [7-9], Ga<sub>2</sub>O<sub>3</sub>-based gate dielectrics are of significant importance to be achieved and to outperform the

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conventional silicon technology, due to their excellent stability and relatively high dielectric constant (approximately 14.2) as compared to that of SiO<sub>2</sub> (approximately 3.9) or even the typically used high- $\kappa$  Al<sub>2</sub>O<sub>3</sub> (approximately 8) [1,10].

Till now, there are several effective integrations of  $Ga_2O_3$ -based gate dielectrics demonstrated in thin-film III-V field-effect transistors (FETs). For instance,  $Ga_2O_3$  and  $Ga_2O_3/Gd_2O_3$  composite materials have been shown to be excellent gate dielectrics for GaAs,  $In_xGa_{1-x}As$ , and GaN thin-film transistors with the low interface state density and high breakdown field strength [2,3,7,11]. However, there are still very few studies focused on  $Ga_2O_3$  dielectrics prepared directly on III-V NWs since the typical thermal oxidizing method is challenging to be executed on the small-diameter NWs, while the atomic-layer-deposited (ALD) high- $\kappa$  HfO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> dielectrics often have significant interfacial defects when performed on NW materials [12]. In this case, it is necessary to explore

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other alternative dielectrics such as  $Ga_2O_3$  achieved by other advanced techniques in order to tackle this issue for the versatile high-mobility III-V NW devices.

Among many  $Ga_2O_3$  phases, the monoclinic  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> is the most stable phase, being a promising gate dielectric alternative; nevertheless, it often requires synthesis at high temperatures to maintain its excellent crystallinity. For example,  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs are usually prepared at above 1,000°C, employing Ga metal as the source in the chemical vapor deposition (CVD) [13], and sometimes even high-energy arc plasma is utilized when using GaN as the starting material [14]. As most III-V NWs are synthesized at a moderate temperature in the range 400°C to 600°C via vapor-liquid-solid (VLS) and/or vaporsolid-solid (VSS) mechanisms [15-18], a compatible low-temperature  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> growth technique is therefore essential to grow dielectrics laterally on III-V NWs while not degrading the III-V NW materials with high vapor pressures.

Recently, we have adopted various III-V material powders as precursor sources for the NW growth by CVD, such as obtaining GaAs, InP, GaSb, etc. at a temperature of 500°C to 600°C [19-21]. Here, in this report, we perform detailed studies on the synthesis behaviors and fundamental physical properties of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs at this moderate growth temperature in a similar CVD growth system. It is revealed that highly crystalline and insulating  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs are successfully grown on the amorphous SiO<sub>2</sub> substrate, which provides a preliminary understanding of the  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs attained by the solid-source CVD method, and further enables us to manipulate the process parameters to achieve high-quality gate dielectrics laterally grown on III-V semiconductor NWs for the coaxially gated NW device structures [22].

# Methods

# Synthesis of Ga<sub>2</sub>O<sub>3</sub> NWs

The Ga<sub>2</sub>O<sub>3</sub> NWs were synthesized in a dual-zone horizontal tube furnace, where the upstream zone was used for evaporating the solid source and the downstream zone for the NW growth, as reported previously [15]. At first, 50nm Au colloids (standard deviation of approximately 5 nm, NanoSeedz, Hong Kong) were drop-casted on SiO<sub>2</sub>/Si substrates (50-nm thermally grown oxide) to serve as the catalyst, which were then placed in the middle of the downstream zone with a tilt angle of approximately 20°. The solid source, GaAs powders (approximately 1.0 g), was contained in a boron nitride crucible, which was then positioned in the upstream zone with a distance of approximately 10 cm away from the substrate with catalysts. During the NW growth, the substrate was initially heated to the preset growth temperature (580°C to 620°C) and the source was then heated to the required source temperature (900°C). Mixture of argon (Ar, 99.9995% purity, 100 sccm) and oxygen (O2, 99.9995% purity) in different flow ratios (100:1 to 100:100) was used as the carrier gas to transport the thermally vaporized precursors to the downstream. After the growth of 1 h, the source and substrate heater were stopped together and cooled down to room temperature under the Ar and  $O_2$  flow.



## Characterization of Ga<sub>2</sub>O<sub>3</sub> NWs

Surface morphologies of the grown Ga<sub>2</sub>O<sub>3</sub> NWs were examined with a scanning electron microscope (SEM; FEI/ Philips XL30, Hillsboro, OR, USA) and transmission electron microscope (TEM; Philips CM-20, Amsterdam, The Netherlands). Crystal structures were determined by collecting X-ray diffraction (XRD) patterns on a Philips powder diffractometer using Cu K $\alpha$  radiation ( $\lambda = 1.5406$  Å) and by selected area electron diffraction (SAED; Philips CM-20). Elemental analysis was performed using an energy-dispersive X-ray (EDS) detector attached to JEOL CM-20 (Akishima-shi, Japan) to measure the chemical composition of the grown NWs. For the TEM and EDS analyses, the Ga<sub>2</sub>O<sub>3</sub> NWs were first suspended in an ethanol solution by ultrasonication and drop-casted onto a copper grid for the corresponding characterization. The reflectance spectrum was measured with a LAMBDA 750 spectrophotometer (PerkinElmer, Waltham, MA, USA) at room temperature.

The Ga<sub>2</sub>O<sub>3</sub> NW arrays were fabricated by contact printing on SiO<sub>2</sub>/Si substrates (50-nm thermally grown oxide) as reported previously [23]. Typically, a pre-patterned SiO<sub>2</sub>/ Si substrate coated with a photoresist was used as the receiver, while the donor NW chip was flipped onto the receiver and slid at a rate of 10 mm/min with a pressure of 50 g/cm<sup>2</sup>. After photoresist removal, the Ga<sub>2</sub>O<sub>3</sub> NW arrays were left on the patterned region. Then, photolithography was utilized to define the electrode regions, and a 100-nmthick Ni film was thermally deposited as the contact electrode followed by a lift-off process. The electrical performance of the fabricated NW arrays was characterized with a standard electrical probe station and Agilent 4155C semiconductor analyzer (Santa Clara, CA, USA).

# **Results and discussion**

As reported previously, we synthesized GaAs NWs by the solid-source CVD method using GaAs powders as the source material heated at 900°C and 100-sccm  $H_2$  as the carrier gas, catalyzed by Au nanoparticles at 580°C to 620°C [15,24]. In an attempt to prepare Ga<sub>2</sub>O<sub>3</sub> in a compatible circumstance, we employ the same conditions here except the  $\mathrm{H}_2$  carrier gas, which is substituted by a mixture of Ar and  $O_2$  in order to introduce oxygen into the growth environment. The flow rate of Ar is fixed at 100 sccm, and the ratio of Ar and  $O_2$  is precisely tailored as 100:1, 100:2, 100:10, and 100:100 by using mass flow controllers (the pressure is approximately 0.8, approximately 0.8, approximately 0.9, and approximately 1.4 Torr, respectively). The corresponding obtained NW products appeared whitish on the substrate, in contrast with the yellowishgreen GaAs NWs. The NWs are then observed by SEM as shown in Figure 1a,b,c,d. It is clear that the NWs grown at the Ar:O<sub>2</sub> flow ratio of 100:2 are relatively long and smooth on the surface (Figure 1b), while the lower  $O_2$  flow induces a significant coating problem (Figure 1a) and the higher  $O_2$ flow suppresses the NW growth (Figure 1c,d). The high  $O_2$ flow might deactivate the Au catalyst leading to no NW growth, while the low O<sub>2</sub> flow might not make the Ga<sub>2</sub>O<sub>3</sub> NW nucleation sufficient over the GaAs NW growth but only overcoat on the GaAs NW surface resulting in the overcoating problem. Notably, in our former study of GaAs NWs, the GaAs powder source has depleted less than 0.1 g of weight after the growth, whereas the source has now depleted more than 0.5 g of weight in this Ga<sub>2</sub>O<sub>3</sub> NW growth by introducing a small amount of oxygen. This would be attributed to the fact that even though Ga has a decently high vapor pressure, there is still a small amount of Ga being evaporated and transported in the H<sub>2</sub> atmosphere in the GaAs NW growth. On the other hand, when  $O_2$  is introduced in the  $\mathrm{Ga}_2\mathrm{O}_3$  NW growth, Ga is easily oxidized to Ga<sub>2</sub>O [25], which has a far higher vapor pressure than that of metallic Ga, and thus can be massively evaporated and





transported by the carrier gas to the substrate; as a result, a proper control in the amount of  $O_2$  feed is critical for the effective NW growth here.

The NWs grown at the Ar:O<sub>2</sub> flow ratio of 100:2 are then observed by TEM as depicted in Figure 2a, which further confirms the straight NWs with smooth surfaces. Furthermore, the elemental composition is analyzed by EDS, and the typical spectrum is illustrated in Figure 2b, which clearly demonstrates that the NWs are mainly composed of Ga and O with an atomic ratio of approximately 2:3. These results evidently show that the obtained NWs here are Ga<sub>2</sub>O<sub>3</sub> instead of the GaAs NWs grown in the H<sub>2</sub> atmosphere. It should also be noted that although Asdoped In<sub>2</sub>O<sub>3</sub> NWs were prepared in a similar system when utilizing InAs powders as the source material and As is detected in the EDS spectrum [26], no As-related signal is obtained within the detection limit of EDS performed in this study. This difference may be due to the alteration in the synthesis condition that  $H_2$  is intentionally introduced into the Ar/O2 carrier gas to suppress the oxide growth in [25], which can be ruled out in this Ga<sub>2</sub>O<sub>3</sub> NW growth. It is plausible that since oxygen has a far higher electron negativity (approximately 3.44) than arsenic (approximately 2.18) and that Ga<sub>2</sub>O<sub>3</sub> has a far lower Gibbs free energy (approximately -998.3 kJ/mol) than GaAs (-67.8 kJ/mol) [27], in this case, Ga<sub>2</sub>O<sub>3</sub> is more preferentially grown from the thermal dynamics point of view. In other words, when  $H_2$  in introduced,  $Ga_2O_3$ growth would be deterred and get substituted by the GaAs growth [25].

In order to investigate the crystal structure of the obtained Ga<sub>2</sub>O<sub>3</sub> NWs, the XRD pattern is attained for NWs readily grown on the SiO<sub>2</sub>/Si substrate as presented in Figure 3a. It is obvious that the NWs are grown in the monoclinic structure ( $\beta$ -phase) in accordance with the standard card PDF 011-0370. Then, the crystal structure and growth orientation of individual NWs are further studied by using SAED as shown in Figure 3b,c,d. All these indicate that the representative NWs all existed in the monoclinic crystal structure, which is in good agreement with the XRD results. Even though the orientations are



observed to vary from NW to NW, typically low-index directions such as [100],  $[\bar{1}11]$ , and  $[\bar{2}13]$  are perceived, which might have resulted from the similar surface energies of these crystal planes, especially for materials in the nanometer size with the examples reported in Si NWs [28], GaAs NWs [15], ZnSe NWs [29], etc.

The bandgap of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs can also be determined by the reflectance spectrum as depicted in Figure 4. It clearly shows that the absorption edge lies at approximately 251 nm (4.94 eV). This bandgap value is in good agreement with that of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs reported in the literature (approximately 254 nm) [30] while a bit higher than that of bulk materials (approximately 270 nm) [31]. A relatively larger bandgap of nanomaterials is often observed than their bulk counterparts, which is usually attributed to the quantum confinement effect of nanomaterials, inducing a blueshift of the bandgap [32].

To shed light on exploring the electronic properties of achieved  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs, the resistance of NWs is first assessed by defining electrodes by standard photolithography. It should be noted that when defining Ni electrodes on a single  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NW, no significant current can be obtained as compared with the resolution (approximately 1 pA) of our semiconductor analyzer and probe station. In order to enlarge the current signal to a measurable level, the  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs are then aligned into parallel arrays by the contact printing technique as reported previously [8,23]. Ni electrodes (with the work function of approximately 5.1 eV) are then defined on both ends of the NW arrays, given in the SEM image in Figure 5a. The NW density is approximately 1 NW/µm, accounting for approximately 200 NWs in the 200 µm (width)  $\times 3.7 \ \mu m$  (length) channel area. In this way, the current-voltage (*I-V*) curve of the representative  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NW array is measured and shown in Figure 5b, where the resistance is estimated to be approximately  $2 \times 10^{12} \Omega$  as the current is approximately 5 pA under 10-V bias. As a result, the resistance is approximately  $4 \times 10^{14} \Omega$  per individual NW (approximately  $2 \times 10^{12} \times 200 \Omega$ , as 200 NWs are connected in parallel). Then, the resistivity can be estimated as  $2 \times 10^{12} \times 200 \ \Omega \times 3.7 \ \mu m/3.14/50^2 \ nm^2 = 8.5 \times 10^7 \ \Omega$ cm, considering the NW diameter of approximately 100 nm. Notably, other metal electrodes with different work functions such as Al (approximately 4.2 eV) and Au (approximately 5.3 eV) are also prepared, in which the results attained are all similar as shown in Figure 5b, suggesting the highly insulating property of the NWs here. This resistivity is relatively larger than those of doped and undoped  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs reported in the literature [4,6,13], which can be attributed to the moderate growth temperature employed in this work such that less impurity would be incorporated, showing its prospective in dielectric materials for advanced III-V nanowire-based nanoelectronics.



# Conclusions

Highly crystalline  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs are synthesized by a solid-source chemical vapor deposition method employing GaAs powders as the source material and mixture of Ar and O<sub>2</sub> as the carrier gas. The NWs grown at the Ar:O<sub>2</sub> flow ratio of 100:2 are long (>10  $\mu$ m) with a uniform diameter of approximately 100 nm and smooth surfaces. X-ray diffraction and selected area electron diffraction results confirm the monoclinic structure of the obtained NWs with varied growth orientations along the low-index planes. Furthermore, the reflectance spectrum demonstrates the bandgap of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> NWs being 4.94 eV, while the electrical measurement deduces the corresponding resistivity of  $8.5 \times 10^7 \ \Omega \ cm$ . All these results indicate the successful synthesis of a large-bandgap Ga2O3 material in III-V-compatible growth conditions, illustrating the promising potential for dielectric materials used for III-V nanowire-based metal-oxide-semiconductor technology.

#### **Competing interests**

The authors declare that they have no competing interests.

#### Authors' contributions

NH synthesized the  $Ga_2O_3$  NWs and drafted the manuscript. FW made the SEM and TEM observations, ZY carried out the XRD measurement, and SY carried out the reflectance spectrum. GD fabricated the NW array devices, and HL made the *I-V* measurement. MF made the SAED identification, and TH carried out the EDS spectrum. JCH provided the idea and completed the manuscript. All authors read and approved the final manuscript.

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