Towards high-mobility In_{2x}Ga_{2-2x}O₃ nanowire field-effect transistors

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ABSTRACT

Recently, owing to the excellent electrical and optical properties, n-type In₂O₃ nanowires (NWs) have attracted tremendous attention for application in memory devices, solar cells, and ultra-violet photodetectors. However, the relatively low electron mobility of In₂O₃ NWs grown by chemical vapor deposition (CVD) has limited their further utilization. In this study, utilizing *in-situ* Ga alloying, highly crystalline, uniform, and thin $In_{2x}Ga_{2-2x}O_3$ NWs with diameters down to 30 nm were successfully prepared via ambient-pressure CVD. Introducing an optimal amount of Ga (10 at.%) into the In₂O₃ lattice was found to effectively enhance the crystal quality and reduce the number of oxygen vacancies in the NWs. A further increase in the Ga concentration adversely induced the formation of a resistive β -Ga₂O₃ phase, thereby deteriorating the electrical properties of the NWs. Importantly, when configured into global back-gated NW field-effect transistors, the optimized $In_{18}Ga_{0.2}O_3$ NWs exhibit significantly enhanced electron mobility reaching up to 750 cm²·V⁻¹·s⁻¹ as compared with that of the pure In_2O_3 NW, which can be attributed to the reduction in the number of oxygen vacancies and ionized impurity scattering centers. Highly ordered NW parallel arrayed devices were also fabricated to demonstrate the versatility and potency of these NWs for next-generation, large-scale, and high-performance nanoelectronics, sensors, etc.

1 Introduction

In the past decades, because of the excellent chemical and physical properties, wide band gap oxide semiconductors have attracted extensive interest and are recognized as promising fundamental building blocks for next-generation electronics, gas sensors, optoelectronics, and others [1–9]. Particularly, as a

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technologically important n-type oxide semiconductor, more recent attention has been focused on indium oxide (In₂O₃) nanowire (NW) materials as compared with other oxide counterparts because of their unique electronic properties. For instance, fully transparent transistors based on single crystalline In₂O₃ NWs synthesized by pulsed laser deposition have been demonstrated with excellent field-effect electron mobilities (μ_{e}), up to ~ 512 cm²·V⁻¹·s⁻¹ [10, 11]. However, although chemical vapor deposition (CVD) is widely recognized as an effective method for the large-scale and low-cost synthesis of nanomaterials, the transistor performance of In₂O₃ NWs enabled by ambient-pressure CVD via vapor-liquid-solid (VLS) growth mechanism still remains insufficient $(\mu_e \sim 200 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1})$ for advanced applications [12–15]. This poor performance is mainly attributed to the uncontrollable radial growth (e.g., overgrowth) leading to the growth of thick and non-uniform In₂O₃ NWs induced by the well-known Gibbs-Thomson effect, especially, the observation of unfavorable nanostructures such as zig-zag, nanocrystal chains, and networks of NWs [16-22]. Furthermore, there are always a large number of oxygen vacancies within the In₂O₃ NWs and these oxygen vacancies could act as impurity scatter centers, thereby deteriorating the transport of carriers and hindering the electrical performance of the subsequently fabricated NW devices [14, 20, 23]. Therefore, it is challenging to achieve high-quality thin In2O3 NWs with controllable diameters and less oxygen vacancies via the CVD growth.

At the same time, previous works have reported the key role of Ga alloying or doping in the synthesis process and its influence on the electrical properties of indium gallium zinc oxide thin-films and zinc oxide NW-based transistors [24–27]. Ga alloying is shown to effectively enable the synthesis of thin and ultra-thin oxide semiconductors to achieve high-quality oxide-lattice structures during their synthesis [28–30]. As compared with the In atoms, the atomic Ga can minimize the formation of intrinsic defects such as oxygen vacancies, because Ga has a relatively higher binding energy with oxygen atom and preferentially forms stronger chemical bonds with it [31, 32]. In this case, the incorporation of Ga into the In_2O_3 lattice is anticipated to improve the bonding of oxygen with cations and correspondingly reduce the defect density (e.g., number of oxygen vacancies); however, there are still very limited studies on exploring reliable Ga alloying to achieve $In_{2x}Ga_{2-2x}O_3$ nanostructures, which may provide important information for achieving further improvement in the device performance.

Here, we introduce controllable *in-situ* Ga alloying into In₂O₃ to attain crystalline and uniform In_{2x}Ga_{2-2x}O₃ NWs with diameters in the range of 20-40 nm via ambient-pressure CVD. In specific, the incorporation of Ga atoms into the In₂O₃ lattice is found to suppress the uneven radial NW growth, to improve the NW crystallinity and more importantly to reduce the formation of oxygen vacancies. With optimal Ga alloying (10 at.%), when configured into field-effect transistors (FETs), the single $In_{1.8}Ga_{0.2}O_3$ NW and the parallel NW arrays exhibit much improved peak electron mobilities with maximum values of up to 750 and 210 cm²·V⁻¹·s⁻¹, respectively. The device threshold voltage (V_{th}) is also observed to shift towards the positive direction with increasing Ga concentration, which indicates the slight decrease in the free electron concentration, being consistent with the reduction in the donor-like oxygen vacancies. All these results would provide valuable insights for achieving high-performance In₂O₃ NWs by *in-situ* Ga alloying for advanced electronic and optoelectronic applications.

2 **Experimental section**

2.1 Nanowire synthesis

Si/SiO₂ (50-nm-thick thermally grown oxide) substrates with a layer of 0.1 nm (nominal thickness) Au film pre-deposited by thermal evaporation was used for the growth of In_2O_3 and $In_{2x}Ga_{2-2x}O_3$ NWs by the ambient-pressure CVD method. For the growth of In_2O_3 NWs, 1.5 g of high-purity In metal granules (1–2 mm in size; 99.999%; China Rare Metal) was used as the In source. For the growth of $In_{2x}Ga_{2-2x}O_3$ NWs, the metal granules of Ga and In (1–2 mm in size; 99.999%; China Rare Metal) were used at different molar ratios (e.g., 3:1 for $In_{1.8}Ga_{0.2}O_3$, 1:1 for $In_{1.52}Ga_{0.48}O_3$, and 1:3 for $In_{0.66}Ga_{1.34}O_3$ NWs; 1.5 g of the source materials in total for each growth) in order to control the stoichiometry of the NWs. Further, 0.5 g of a graphite powder (< 20 µm in size, synthetic; Sigma-Aldrich) was mixed with the metal source as the precursor. During the growth, the precursor mixture was placed at the sealed end whereas the substrate was positioned at the open end of a small quartz tube (10 cm in length and 1 cm in diameter). The entire set-up was then placed in the center of a large quartz tube (1 inch in diameter), located in a single-zone horizontal tube furnace. Next, a mixed gas of 4% oxygen (99.999%) and 96% argon (99.9995%) was introduced as the carrier gas into the quartz tube. The furnace was heated to 1,030 °C in 30 min, held for 30 min, and then cooled to room temperature naturally. After the synthesis, a fluffy white product was found on the substrate. The growth was also noted to be highly sensitive to the temperature. When the temperature was set below 1,030 °C, no NW was obtained. Once the temperature was adjusted to >1,030 °C, significant coatings were observed on the NWs.

2.2 Material characterization

The crystal structures of the obtained NWs were determined by X-ray diffraction (XRD) performed on a powder X-ray diffractometer (RIGAKU, Philips) using Cu K α radiation (λ = 1.5406 Å). The NW morphologies were examined using a scanning electron microscope (SEM, XL30, FEI) and transmission electron microscope (TEM, Tecnai F30, FEI). Highresolution TEM (HRTEM) observations and selected area electron diffraction (SAED) were also performed to assess the crystal structure and crystallinity of the NWs. For the TEM studies, the NWs were first suspended in ethanol by ultrasonication and then drop-casted onto TEM Cu grids. The chemical composition of the NWs was determined using an energy dispersive X-ray spectrometer (EDS) attached to the SEM and TEM. The corresponding EDS mappings were also obtained from EDS attached to TEM.

2.3 Single NW FET fabrication and electrical measurement

After the growth, the NWs were drop-casted onto degenerately boron-doped Si substrates with a

50-nm-thick thermally grown gate oxide layer on top. Photolithography was utilized to define the source and drain regions, and 5/80-nm thick Ti/Au film was deposited by electron beam evaporation followed by a lift-off process to serve as the contact electrodes. The electrical performance of the fabricated back-gate FETs was then evaluated using a standard electrical probe station with an Agilent 4155C semiconductor analyzer (Agilent Technologies, Santa Clara, CA, USA).

3 Results and discussion

As shown in the SEM images in Figs. 1(a) and 1(b), both In_2O_3 and Ga-alloyed In_2O_3 (e.g., $In_{1.8}Ga_{0.2}O_3$) products have wire-like morphology with lengths of tens of micrometers. However, the distribution of the diameters of the NWs in the two samples is very different although the same catalyst film thickness was utilized for the growth. The NW diameter is found to be significantly reduced after Ga alloying. In order to understand the difference, statistical distributions of the NW diameters of In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ were evaluated and the results are presented in Fig. 1(c); the diameters were extracted from the TEM images of the NWs. It is clear that the diameter of In_2O_3 NWs is 96 ± 23 nm, whereas the



Figure 1 Morphology, diameter distribution, and crystal structure of as-prepared In_2O_3 NWs and $In_{1.8}Ga_{0.2}O_3$ NWs. (a) SEM image of In_2O_3 NWs. (b) SEM image of $In_{1.8}Ga_{0.2}O_3$ NWs. (c) Histogram for the diameter distribution with Gaussian fitting performed. (d) XRD patterns of the obtained NWs.

diameter of $In_{1.8}Ga_{0.2}O_3$ NWs is 32 ± 12 nm, which is ~ 3 times smaller than that of pure In₂O₃ NWs. Further, once Ga atoms are introduced into the In₂O₃ lattice, the NW diameter is decreased substantially and remains relatively constant as the Ga concentration is increased further up to 67 at.% (data not shown). It is also noted that, on average, the In_{1.8}Ga_{0.2}O₃ NWs are much longer than the In₂O₃ NWs. Specifically, the longitudinal growth rates of In₂O₃ and In₁₈Ga_{0.2}O₃ NWs were qualitatively estimated as 0.9 and 2.1 µm·min⁻¹, respectively (Fig. S1 in the Electronic Supplementary Material (ESM)). The presence of Ga would typically decrease the vapor pressure of the corresponding metal oxides, leading to a higher longitudinal growth rate [33, 34]. Based on the well-known Gibbs-Thomson effect, the observed increase in the longitudinal growth rate usually leads to smaller diameters of the NWs grown via the VLS mechanism [35, 36]. All these results evidently indicate that Ga alloying can effectively prevent the radial growth to yield thin and uniform NWs. At the same time, EDS was performed to assess the chemical composition of the obtained NWs. Figure S2 in the ESM displays a typical EDS spectrum of NWs grown with the specific precursor mixing ratio of 3:1 (In:Ga), in which the In, Ga, and O signals are clearly observed with the elemental ratio of 1:0.11:2.31, indicating the stoichiometry of the NW being $In_{1.8}Ga_{0.2}O_3$ with 10 at.% Ga alloying. XRD was then carried out to evaluate the crystal structure of the NWs. As depicted in Fig. 1(d), all the diffraction peaks can be indexed to cubic In₂O₃ (JCPDS Card No. 06-0416), suggesting the phase purity of In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs [18, 37]. The high phase purity suggests the absence of any other impurity phase. In the high-resolution XRD patterns (Fig. S3 in the ESM), the (400) peak of In_2O_3 is observed to shift to higher angles, from 35.50° to 35.57°, after Ga alloying, indicating the incorporation of Ga into the lattice of In₂O₃. However, upon increasing the Ga concentration further to more than 24 at.%, additional peaks associated with β -Ga₂O₃ start to appear, revealing the phase segregation between In_2O_3 and β -Ga₂O₃ within the NWs (Fig. S4 in the ESM).

In addition, the surface morphology of the obtained NWs could be confirmed by detailed TEM characterization, as illustrated in Figs. 2(a) and 2(b).

It is obvious that the typical diameter of In₂O₃ NWs is 80 nm, whereas the representative diameter of In_{1.8}Ga_{0.2}O₃ NWs is 28 nm. The reduced NW diameter after Ga alloying is perfectly consistent with the SEM results. The single-crystalline nature of the In₂O₃ and $In_{18}Ga_{02}O_3$ NWs could also be confirmed by the SAED patterns (see the inset of Figs. 2(a) and 2(b), respectively). From the SAED patterns, the growth direction of both In₂O₃ and In_{1.8}Ga_{0.2}O₃ NWs is found to be the same, which is along the [110] direction [11, 16], suggesting that Ga alloying up to 10 at.% does not have any noticeable impact on the lattice structure owing to the small difference between the atomic radius of Ga and In as well as the high compatibility of Ga atoms with the cubic structure of In₂O₃. The excellent crystallinity of the NW is also evidenced by the clear lattice fringes shown in Figs. 2(c) and 2(d). In specific, the lattice spacing of In_2O_3 is measured to be 0.255 nm, corresponding to the {400} planes of cubic In_2O_3 . The lattice spacing of $In_{18}Ga_{0.2}O_3$ for the same plane is slightly smaller at 0.243 nm,



Figure 2 TEM characterization and elemental analysis. (a) and (b) TEM images and the corresponding SAED patterns with the zone axis of [001] (inset) of representative In_2O_3 NWs and $In_{1.8}Ga_{0.2}O_3$ NWs, respectively. (c) and (d) HRTEM images of In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs, respectively. (e) Scanning TEM image of the $In_{1.8}Ga_{0.2}O_3$ NW with the elemental mapping of Ga, In, and O, respectively.

because of the smaller atomic radius of Ga than that of In [24]. To investigate the uniformity of the distribution of the constituents in $In_{1.8}Ga_{0.2}O_3$ NWs, the corresponding EDS mapping was performed (Fig. 2(e)). It is evident that Ga, In, and O are homogenously distributed along the NW. All these results suggest that Ga is uniformly distributed in the In_2O_3 lattice, without any significant segregation and defect formation.

Apart from the excellent crystallinity and uniform elemental distribution, it is important to determine the electrical property of these $In_{2x}Ga_{2-2x}O_3$ NWs. In this study, single NW and parallel NW array-based devices configured in the global back-gated geometry were employed; an illustrative schematic of the device and the corresponding SEM image are shown in the inset of Fig. 3(a). Based on the typical transfer characteristics of both single In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NW devices (Fig. 3(a)), it is clear that the sourcedrain current (I_{ds}) increases with increasing back-gate voltage (V_{gs}), indicating the n-type conducting behaviors of both types of NWs. In specific, the In₂O₃ and In_{1.8}Ga_{0.2}O₃ NWFETs deliver ~ 25 and ~ 30 μ A on-current under V_{ds} = 1 V and V_{gs} = 20 V, respectively. The hysteresis in the transfer characteristics is possibly induced by the contamination of the nanowire surface or the gate oxide in the proximity of the nanowire, moveable ions in the dielectric, absorbents from air, etc. Further, the linear I_{ds} - V_{ds} relationship in the output characteristics further confirms the nearly ohmic contact between the NWs and the electrodes (Fig. 3(b)). The free carrier concentration (*n*) of the as-synthesized NWs can be extracted from the transfer curves in Fig. 3(a) using Eq. (1)

$$n = \left| \frac{4C_{\rm ox} V_{\rm th}}{\pi q d^2 L} \right| \tag{1}$$

where, C_{ox} is the gate capacitance that can be accurately obtained from the finite element analysis software COMSOL, V_{th} is the threshold voltage, q is the charge



Figure 3 Electrical characterization of typical FETs fabricated with In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs. (a) Transfer characteristics of two representative back-gated FETs based on single In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs, respectively ($V_{ds} = 1$ V). The inset shows the SEM image of the FET with 3-µm channel length fabricated with the $In_{1.8}Ga_{0.2}O_3$ NW and the schematic illustration of the device. The scale bar is 1 µm. (b) Corresponding output characteristics of the NWFET under $V_{gs} = -10$, -7, -4, -1, 2, and 5 V from bottom to top. (c) Field-effect electron mobility assessment for the same set of FETs presented in (a) with $V_{ds} = 1$ V. (d) Statistical results of the peak field-effect electron mobility of ~ 80 FETs based on In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs. Gaussian fitting is performed on the histogram.

of an electron, *d* is the diameter of the NW, and *L* is the length of the FET channel [21, 23]. As compared with the device based on the In₂O₃ NW channel, the In_{1.8}Ga_{0.2}O₃ NWFET exhibits a lower carrier concentration at the zero gate voltage and the *n* value decreases from 1.21×10^{19} to 1.05×10^{19} cm⁻³, accordingly. At the same time, the μ_e of single NW FET can also be calculated from the transfer curves using Eq. (2), as presented in Fig. 3(c) [38, 39].

$$\mu_{\rm e} = \frac{g_{\rm m} L^2}{C_{\rm ox} V_{\rm ds}} \tag{2}$$

Here, g_m is the transconductance, which is dI_{ds}/dV_{gs} . Thus, the peak μ_e values of the In₂O₃ and In_{1.8}Ga_{0.2}O₃ NW devices are found to be 320 and 700 cm²·V⁻¹·S⁻¹, respectively. It is obvious that the electron mobility of the In_{1.8}Ga_{0.2}O₃ NW FET is more than twice of that of pure In₂O₃ NW, indicating that Ga alloying at 10 at.% is beneficial for enhancing the performance of the In₂O₃ NW electronic device. To further confirm this trend, statistical distributions of the mobilities of more than 30 In₂O₃ and In_{1.8}Ga_{0.2}O₃ NW devices with similar NW diameters were collected for each sample group, as depicted Fig. 3(d). Evidently, the average mobility (590 cm²·V⁻¹·s⁻¹) of $In_{1.8}Ga_{0.2}O_3$ NW devices is more than twice that of pure In_2O_3 NW devices (210 cm²·V⁻¹·s⁻¹), confirming the enhancement of the electron mobility simply through Ga alloying. The slight variation between the peak mobility and average mobility can be attributed to the dependence of the electrical properties on the NW crystallinity, surface roughness, and many other factors.

To shed light on the enhancement of the electron mobility of In_{1.8}Ga_{0.2}O₃ NW devices, X-ray photoelectron spectroscopy (XPS) was carried out. As shown in Figs. 4(a) and 4(b), the In 4d peak and the O 1s peaks of pure In₂O₃ NWs are clearly observed in the XPS spectra. For In_{1.8}Ga_{0.2}O₃ NWs, apart from the In 4d and O 1s peaks, the Ga 3d peak is also observed (Fig. 4(c)), indicating the incorporation of Ga in the In₂O₃ NWs. Further, there is a tiny shift of the In 4d peak as compared with that of pure In₂O₃ NWs, which may be attributed to the size effect, as reported in the literature [40, 41]. Interestingly, the XPS peak profile of O 1s of In_{1.8}Ga_{0.2}O₃ NWs is substantially different from that of pure In₂O₃ NWs, as shown in Figs. 4(b) and 4(d). Specifically, Gaussian fittings of the peak profiles were made to clearly illustrate the



Figure 4 XPS characterization of the obtained NWs. XPS spectra of the In 4d, Ga 3d, and O 1s levels of In_2O_3 NWs (a) and (b) and $In_{1.8}Ga_{0.2}O_3$ NWs (c) and (d). Gaussian-resolved fitting curves labeled as O_L , O_M , and O_H , designating the O^{2^-} ions in the cubic structure, O^{2^-} ions in the oxygen deficient region, and loosely bound oxygen atoms on the surface, respectively.

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observed difference. The component labeled as O_H located at 532.25 \pm 0.2 eV, with a higher binding energy, is usually attributed to the presence of loosely bound oxygen on the surface of NWs, belonging to some chemisorbed species such as adsorbed H₂O or O_2 . The medium binding energy component at 530.8 ± 0.15 eV, labeled as O_{M} is related to the O²⁻ ions that are in the oxygen-deficient regions within the In₂O₃ matrix [30, 42, 43]. Therefore, the changes in the intensity of this component are usually associated with the variation in the concentration of the oxygen defects. The component at the lower binding energy $(529.6 \pm 0.2 \text{ eV})$, labeled as O_L, represents the O²⁻ ions located in the substitutional sites of the In₂O₃ lattice; these are surrounded by In ions with their full complement of the nearest neighbor O²⁻ ions. In other words, the intensity of this component can simply be a measure of the number of oxygen atoms located in the substitutional sites, with fully oxidized stoichiometry. At the same time, to further analyze the relationship between different components of the O 1s peak, the area ratio of O_L/O_M in In_2O_3 and $In_{1.8}Ga_{0.2}O_3$ NWs were determined to be 1.4 and 2.6, respectively. This result indicates that the number of oxygen vacancies of In_{1.8}Ga_{0.2}O₃ NWs is qualitatively smaller than that of pure In₂O₃ NWs. Since the Ga–O bonds are typically stronger than the In–O bonds in terms of their bond energies, the optimal Ga alloying in In₂O₃ NWs can improve the effectiveness of oxygen bonding with metal constituents and hence decreases the density of oxygen vacancies there. Explicitly, the free electron generation can be closely related to the substitution of oxygen according to the following equation [28].

$$O_{o}^{x} = \frac{1}{2}O_{2} + V_{o}^{\bullet \bullet} + 2e^{-}$$
 (3)

Here, the O_2 species can be formed by O atoms from the oxide sublattice. The doubly positively charged oxygen vacancy and two free electrons are also created in the same process, indicating that the decrease in the free electron concentration is merely a result of the reduction in the number of oxygen vacancies. Furthermore, the oxygen vacancies can as well act as ionized impurity scattering centers and have a predominant effect on the electron mobility of the NWs. As a result, the reduction in the number of oxygen vacancies would lead to a slight decrease in the free electron concentration and enhanced electron mobility of $In_{1.8}Ga_{0.2}O_3$ NWs as compared with those of their pure In_2O_3 counterparts.

In order to demonstrate the potential application of these In_{1.8}Ga_{0.2}O₃ NWs, large-scale NW parallel arrayed FETs were then fabricated via the wellestablished NW contact printing technique (Fig. 5(a)) [19, 44]. It is clear that an impressive on/off current ratio of 10⁵ could be achieved (Fig. 5(b)). The linear relation of the I_{ds} vs. V_{ds} curves also confirms the nearly ohmic contact between the electrodes and the NWs (Fig. 5(c)). Considering the channel width of 100 µm, the device outputs a current density of ~ 10 μ A· μ m⁻¹ under V_{ds} = 2 V and V_{gs} = 30 V. This current density can be further enhanced by increasing the NW print density in the device. Further, the corresponding field-effect electron mobility of the device was also evaluated using Eq. (2). For these NW arrayed devices, the capacitance is calculated by multiplying the electrostatically modeled gate oxide capacitance for a single NW by the number of NWs in the arrayed device. The extracted peak electron mobility is 210 cm²·V⁻¹·s⁻¹ (Fig. 5(d)). All these performance data are comparable or even better than those of other state-of-the-art n-type thin-film oxide transistors reported in the literature [45, 46]. However, as compared with the single In_{1.8}Ga_{0.2}O₃ NW device (Fig. 3(a)), the performance (e.g., output current density, electron mobility, etc.) of the NW parallel arrayed device seems to deteriorate, which is probably due to the inefficient electrical contact between the NWs and electrodes as well as the ineffective gate coupling owing to the slight misalignment of the printed NW arrays. Inevitably, some fragmented wires would also exist within the channel, which increase the parasitic capacitance and deteriorate the device performance. In the future, this performance can be enhanced by down-scaling the channel length, optimizing the NW printing, and adopting a top-gate device structure with high-dielectrics. Moreover, it is also critical to investigate the effect of the Ga concentration on the electrical properties of the $In_{2x}Ga_{2-2x}O_3$ NW devices (Fig. S5 in the ESM). As discussed above, a higher Ga concentration reduced





Figure 5 Electrical characterization of the fabricated $In_{1.8}Ga_{0.2}O_3$ NW parallel arrayed devices. (a) (Top) Typical SEM image and (bottom) illustrative schematic of a contact-printed back-gated NW parallel arrayed FET with Au electrodes. (b) Transfer curves of the device (channel length = 2.5 µm; channel width = 100 µm; V_{ds} = 2 V). (c) Output curves of the device. (d) Corresponding field-effect electron mobility assessed as a function of gate voltage bias at V_{ds} = 2 V.

the free electron concentration, owing to the formation of an additional phase of β -Ga₂O₃ and corresponding phase segregation between In₂O₃ and β -Ga₂O₃ of the NWs. In this case, as anticipated, with an increase in the Ga concentration from 10 at.% to 67 at.%, the average $V_{\rm th}$ value increases from approximately –14 to 4 V, accordingly (Fig. 6(a)). Meanwhile, the output current first increases from 25.4 to 29.3 µA and then drops continuously to 10.1 and 2.5 µA (Fig. 6(b)), while the $I_{\rm on}/I_{\rm off}$ ratio changes with a similar tendency as



Figure 6 Effect of Ga concentration on the electrical properties of $In_{2x}Ga_{2-2x}O_3$ NW devices. The average value of (a) V_{th} , (b) on-current, (c) I_{on}/I_{off} ratio, and (d) peak mobility extracted from more than 40 NW FETs fabricated with pure In_2O_3 and 10 at.%, 24 at.%, and 67 at.% Ga-alloyed In_2O_3 NWs, respectively.

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that of the output current, being significantly improved from 1.2×10^6 to 5.5×10^7 , and then drops continuously to 1.0×10^6 and 2.8×10^5 (Fig. 6(c)). Importantly, owing to the better crystallinity and the reduction of impurity scattering centers induced by Ga alloying, the average value of the peak electron mobility increases from 218 to 582 cm²·V⁻¹·s⁻¹ with an increase in the Ga concentration up to 10 at.%, and then decreases to 257 and 114 cm²·V⁻¹·s⁻¹, with further increase in the Ga concentration (Fig. 6(d)). This decline in μ_e is mainly caused by precipitation or formation of the β -Ga₂O₃ phase, when the Ga concentration reaches the supersaturation limit within the In_2O_3 lattice; the existence of β -Ga₂O₃ domains would be detrimental to the electrical properties of the NWs owing to their poor electrical conductivity. As a result, the enhanced electrical characteristic of In_{2x}Ga_{2-2x}O₃ NWs is achieved at the optimal Ga concentration for various technological applications.

4 Conclusions

Crystalline and uniform In_{2x}Ga_{2-2x}O₃ NWs were successfully synthesized by ambient-pressure CVD. The Ga concentration plays an important role in the synthesis process and has a great impact on the electrical properties of the obtained NWs, owing to the higher binding energy of Ga with oxygen atoms. In specific, incorporating Ga into the In₂O₃ lattice is found to be effective for controlling the non-uniform NW radial growth and minimizing the growth of unfavorable nanostructures of In₂O₃. Furthermore, the XPS data reveals that the optimal Ga concentration of 10 at.% (i.e., In_{1.8}Ga_{0.2}O₃) would lead to the reduction of oxygen vacancies in In₂O₃ NWs. This optimal Ga concentration also yields enhanced peak electron mobility reaching up to 750 cm²·V⁻¹·s⁻¹ when a global back-gated NW device is fabricated and evaluated. When the Ga concentration was increased further, the NW device performance deteriorated, probably owing to the precipitation or formation of the β -Ga₂O₃ phase with higher resistivity. More importantly, as compared with other conventional n-type thin-film oxide transistors, the NW parallel arrayed device exhibits superior performance with peak electron mobility of 210 cm²·V⁻¹·s⁻¹ and current density of 10 µA·µm⁻¹. All these results demonstrate the great potential of these NWs for next-generation, large-scale, and highperformance nanoelectronics, sensors, etc.

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Electronic Supplementary Material: Supplementary material (cross section of SEM of as-prepared In₂O₃ NWs and In_{1.8}Ga_{0.2}O₃ NWs, EDS spectra of the NWs of In_{1.8}Ga_{0.2}O₃ NWs, XRD data of as-prepared In_{1.52}Ga_{0.48}O₃ and In_{0.66}Ga_{1.34}O₃ NWs, and electrical characterization of back-gate FETs based on single In₂O₃, In_{1.8}Ga_{0.2}O₃, In_{11.52}Ga_{0.48}O₃ and In_{0.66}Ga_{1.34}O₃ NWs) is available in the online version of this article at https://doi.org/10.1007/s12274-018-2106-9.

References

- [1] Liu, Q. Z.; Liu, Y. H.; Wu, F. Q.; Cao, X.; Li, Z.; Alharbi, M.; Abbas, A. N.; Amer, M. R.; Zhou, C. W. Highly sensitive and wearable In₂O₃ nanoribbon transistor biosensors with integrated on-chip gate for glucose monitoring in body fluids. *ACS Nano* **2018**, *12*, 1170–1178.
- [2] Meng, M.; Wu, X. L.; Ji, X. L.; Gan, Z. X.; Liu, L. Z.; Shen, J. C.; Chu, P. K. Ultrahigh quantum efficiency photodetector and ultrafast reversible surface wettability transition of square In₂O₃ nanowires. *Nano Res.* **2017**, *10*, 2772–2781.
- [3] Macco, B.; Knoops, H. C. M.; Kessels, W. M. M. Electron scattering and doping mechanisms in solid-phase-crystallized In₂O₃:H prepared by atomic layer deposition. *ACS Appl. Mater. Interfaces* **2015**, *7*, 16723–16729.
- [4] Park, S.; Kim, S.; Sun, G.-J.; Lee, C. Synthesis, structure, and ethanol gas sensing properties of In₂O₃ nanorods decorated with Bi₂O₃ nanoparticles. *ACS Appl. Mater. Interfaces* 2015, 7, 8138–8146.
- [5] Liu, A.; Liu, G. X.; Zhu, H. H.; Xu, F.; Fortunato, E.; Martins, R.; Shan, F. K. Fully solution-processed low-voltage aqueous In₂O₃ thin-film transistors using an ultrathin ZrO_x

dielectric. ACS Appl. Mater. Interfaces 2014, 6, 17364-17369.

- [6] Khim, D.; Lin, Y.-H.; Nam, S.; Faber, H.; Tetzner, K.; Li, R. P.; Zhang, Q.; Li, J.; Zhang, X. X.; Anthopoulos, T. D. Modulation-doped In₂O₃/ZnO heterojunction transistors processed from solution. *Adv. Mater.* **2017**, *29*, 1605837.
- [7] Leppäniemi, J.; Huttunen, O.-H.; Majumdar, H.; Alastalo,
 A. Flexography-printed In₂O₃ semiconductor layers for high-mobility thin-film transistors on flexible plastic substrate. *Adv. Mater.* 2015, *27*, 7168–7175.
- [8] Kim, J.; Rim, Y. S.; Chen, H. J.; Cao, H. H.; Nakatsuka, N.; Hinton, H. L.; Zhao, C. Z.; Andrews, A. M.; Yang, Y.; Weiss, P. S. Fabrication of high-performance ultrathin In₂O₃ film field-effect transistors and biosensors using chemical lift-off lithography. *ACS Nano* 2015, *9*, 4572–4582.
- [9] Hou, J. G.; Cao, S. Y.; Sun, Y. Q.; Wu, Y. Z.; Liang, F.; Lin, Z. S.; Sun, L. C. Atomically thin mesoporous In₂O_{3-x}/In₂S₃ lateral heterostructures enabling robust broadband-light photo-electrochemical water splitting. *Adv. Energy Mater.* **2018**, *8*, 1701114.
- [10] Ju, S.; Facchetti, A.; Xuan, Y.; Liu, J.; Ishikawa, F.; Ye, P. D.; Zhou, C. W.; Marks, T. J.; Janes, D. B. Fabrication of fully transparent nanowire transistors for transparent and flexible electronics. *Nat. Nanotechnol.* 2007, *2*, 378–384.
- [11] Li, C.; Zhang, D.; Han, S.; Liu, X.; Tang, T.; Zhou, C. Diameter-controlled growth of single-crystalline In₂O₃ nanowires and their electronic properties. *Adv. Mater.* 2003, *15*, 143–146.
- [12] Zou, X. M.; Liu, X. Q.; Wang, C. L.; Jiang, Y.; Wang, Y.; Xiao, X. H.; Ho, J. C.; Li, J. C.; Jiang, C. Z.; Xiong, Q. H. et al. Controllable electrical properties of metal-doped In₂O₃ nanowires for high-performance enhancement-mode transistors. *ACS Nano* **2013**, *7*, 804–810.
- [13] Shen, G. Z.; Xu, J.; Wang, X. F.; Huang, H. T.; Chen, D. Growth of directly transferable In₂O₃ nanowire mats for transparent thin-film transistor applications. *Adv. Mater.* 2011, 23, 771–775.
- [14] Lei, B.; Li, C.; Zhang, D.; Tang, T.; Zhou, C. Tuning electronic properties of In₂O₃ nanowires by doping control. *Appl. Phys. A* 2004, *79*, 439–442.
- [15] Peng, X. S.; Wang, Y. W.; Zhang, J.; Wang, X. F.; Zhao, L. X.; Meng, G. W.; Zhang, L. D. Large-scale synthesis of In₂O₃ nanowires. *Appl. Phys. A* **2002**, *74*, 437–439.
- [16] Lao, J.; Huang, J.; Wang, D.; Ren, Z. Self-assembled In₂O₃ nanocrystal chains and nanowire networks. *Adv. Mater.* 2004, *16*, 65–69.
- [17] Yan, Y. G.; Zhang, Y.; Zeng, H. B.; Zhang, J. X.; Cao, X. L.; Zhang, L. D. Tunable synthesis of In₂O₃ nanowires, nanoarrows and nanorods. *Nanotechnology* **2007**, *18*, 175601.
- [18] Kam, K. C.; Deepak, F. L.; Cheetham, A. K.; Rao, C. N. R.

In₂O₃ nanowires, nanobouquets and nanotrees. *Chem. Phys. Lett.* **2004**, *397*, 329–334.

- [19] Han, N.; Yang, Z. X.; Wang, F. Y.; Yip, S.; Dong, G. F.; Liang, X. G.; Hung, T.; Chen, Y. F.; Ho, J. C. Modulating the morphology and electrical properties of GaAs nanowires via catalyst stabilization by oxygen. ACS Appl. Mater. Interfaces 2015, 7, 5591–5597.
- [20] Zhang, D. H.; Ma, H. L. Scattering mechanisms of charge carriers in transparent conducting oxide films. *Appl. Phys. A* **1996**, *62*, 487–492.
- [21] Yang, Z.-X.; Yip, S.; Li, D. P.; Han, N.; Dong, G. F.; Liang, X. G.; Shu, L.; Hung, T. F.; Mo, X. L.; Ho, J. C. Approaching the hole mobility limit of GaSb nanowires. ACS Nano 2015, 9, 9268–9275.
- [22] Shen, Y. D.; Chen, R. J.; Yu, X. C.; Wang, Q. J.; Jungjohann, K. L.; Dayeh, S. A.; Wu, T. Gibbs–Thomson effect in planar nanowires: Orientation and doping modulated growth. *Nano Lett.* 2016, *16*, 4158–4165.
- [23] Li, W. Q.; Liao, L.; Xiao, X. H.; Zhao, X. Y.; Dai, Z. G.; Guo, S. S.; Wu, W.; Shi, Y.; Xu, J. X.; Ren, F. et al. Modulating the threshold voltage of oxide nanowire field-effect transistors by a Ga⁺ ion beam. *Nano Res.* **2014**, *7*, 1691–1698.
- [24] Nomura, K.; Ohta, H.; Takagi, A.; Kamiya, T.; Hirano, M.; Hosono, H. Room-temperature fabrication of transparent flexible thin-film transistors using amorphous oxide semiconductors. *Nature* 2004, *432*, 488–492.
- [25] Yuan, G. D.; Zhang, W. J.; Jie, J. S.; Fan, X.; Tang, J. X.; Shafiq, I.; Ye, Z. Z.; Lee, C. S.; Lee, S. T. Tunable n-type conductivity and transport properties of Ga-doped ZnO nanowire arrays. *Adv. Mater.* **2008**, *20*, 168–173.
- [26] Park, W. J.; Shin, H. S.; Ahn, B. D.; Kim, G. H.; Lee, S. M.; Kim, K. H.; Kim, H. J. Investigation on doping dependency of solution-processed Ga-doped ZnO thin film transistor. *Appl. Phys. Lett.* 2008, *93*, 083508.
- [27] Kamiya, T.; Nomura, K.; Hosono, H. Origins of high mobility and low operation voltage of amorphous oxide TFTs: Electronic structure, electron transport, defects and doping. *J. Disp. Technol.* 2009, *5*, 468–483.
- [28] Jeong, S.; Ha, Y. G.; Moon, J.; Facchetti, A.; Marks, T. J. Role of gallium doping in dramatically lowering amorphousoxide processing temperatures for solution-derived indium zinc oxide thin-film transistors. *Adv. Mater.* **2010**, *22*, 1346– 1350.
- [29] Kim, G. H.; Jeong, W. H.; Kim, H. J. Electrical characteristics of solution-processed InGaZnO thin film transistors depending on Ga concentration. *Phys. Status Solidi (a)* **2010**, 207, 1677–1679.
- [30] Noh, H.-K.; Chang, K. J.; Ryu, B.; Lee, W.-J. Electronic structure of oxygen-vacancy defects in amorphous In-Ga-Zn-O

semiconductors. Phys. Rev. B 2011, 84, 115205.

- [31] Yao, J. K.; Xu, N. S.; Deng, S. Z.; Chen, J.; She, J. C.; Shieh, H.-P. D.; Liu, P.-T.; Huang, Y.-P. Electrical and photosensitive characteristics of a-IGZO TFTs related to oxygen vacancy. *IEEE. Trans. Electron Dev.* 2011, 58, 1121–1126.
- [32] Zan, H. W.; Yeh, C. C.; Meng, H. F.; Tsai, C. C.; Chen, L. H. Achieving high field-effect mobility in amorphous indiumgallium-zinc oxide by capping a strong reduction layer. *Adv. Mater.* 2012, *24*, 3509–3514.
- [33] Johnson, M. C.; Aloni, S.; McCready, D. E.; Bourret-Courchesne, E. D. Controlled vapor–liquid–solid growth of indium, gallium, and tin oxide nanowires via chemical vapor transport. *Cryst. Growth Des.* **2006**, *6*, 1936–1941.
- [34] Vomiero, A.; Ferroni, M.; Comini, E.; Faglia, G.; Sberveglieri, G. Insight into the formation mechanism of one-dimensional indium oxide wires. *Cryst. Growth Des.* **2010**, *10*, 140–145.
- [35] Schmidt, V.; Senz, S.; Gösele, U. Diameter dependence of the growth velocity of silicon nanowires synthesized via the vapor-liquid-solid mechanism. *Phys. Rev. B* 2007, 75, 045335.
- [36] Fröberg, L. E.; Seifert, W.; Johansson, J. Diameter-dependent growth rate of InAs nanowires. *Phys. Rev. B* 2007, 76, 153401.
- [37] Gao, T.; Wang, T. H. Catalytic growth of In₂O₃ nanobelts by vapor transport. J. Cryst. Growth 2006, 290, 660–664.
- [38] Yang, Z.-X.; Wang, F. Y.; Han, N.; Lin, H.; Cheung, H.-Y.; Fang, M.; Yip, S.; Hung, T.; Wong, C.-Y.; Ho, J. C. Crystalline GaSb nanowires synthesized on amorphous substrates: From the formation mechanism to p-channel transistor applications. *ACS Appl. Mater. Interfaces* **2013**, *5*, 10946–10952.

- [39] Yang, Z.-X.; Han, N.; Fang, M.; Lin, H.; Cheung, H.-Y.; Yip, S.; Wang, E.-J.; Hung, T.; Wong, C.-Y.; Ho, J. C. Surfactant-assisted chemical vapour deposition of highperformance small-diameter GaSb nanowires. *Nat. Commun.* 2014, *5*, 5249.
- [40] Rao, C. N. R.; Kulkarni, G. U.; Thomas, P. J.; Edwards, P. P. Size-dependent chemistry: Properties of nanocrystals. *Chem.—Eur. J.* 2002, *8*, 28–35.
- [41] Volokitin, Y.; Sinzig, J.; de Jongh, L. J.; Schmid, G.; Vargaftik, M. N.; Moiseevi, I. I. Quantum-size effects in the thermodynamic properties of metallic nanoparticles. *Nature* **1996**, *384*, 621–623.
- [42] Chen, M.; Wang, X.; Yu, Y. H.; Pei, Z. L.; Bai, X. D.; Sun, C.; Huang, R. F.; Wen, L. S. X-ray photoelectron spectroscopy and auger electron spectroscopy studies of Al-doped ZnO films. *Appl. Surf. Sci.* 2000, *158*, 134–140.
- [43] Kumar, V.; Swart, H. C.; Ntwaeaborwa, O. M.; Kroon, R. E.; Terblans, J. J.; Shaat, S. K. K.; Yousif, A.; Duvenhage, M. M. Origin of the red emission in zinc oxide nanophosphors. *Mater. Lett.* 2013, 101, 57–60.
- [44] Yerushalmi, R.; Jacobson, Z. A.; Ho, J. C.; Fan, Z. Y.; Javey, A. Large scale, highly ordered assembly of nanowire parallel arrays by differential roll printing. *Appl. Phys. Lett.* 2007, *91*, 203104.
- [45] Lee, J. S.; Chang, S.; Koo, S.-M.; Lee, S. Y. High-performance a-IGZO TFT with ZrO₂ gate dielectric fabricated at room temperature. *IEEE Electron Dev. Lett.* **2010**, *31*, 225–227.
- [46] Nomura, K.; Takagi, A.; Kamiya, T.; Ohta, H.; Hirano, M.; Hosono, H. Amorphous oxide semiconductors for highperformance flexible thin-film transistors. *Jpn. J. Appl. Phys.* 2006, 45, 4303.